

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE

AD-A 200 108

REPORT DOCUMENTATION PAGE				Form Approved OMB No. 0704-0188 Exp Date Jun 30, 1986	
1a. REPORT SECURITY CLASSIFICATION UNCLASSIFIED			1b. RESTRICTIVE MARKINGS		
2a. SECURITY CLASSIFICATION AUTHORITY			3. DISTRIBUTION/AVAILABILITY OF REPORT Approved for public release; distribution unlimited		
2b. DECLASSIFICATION/DOWNGRADING SCHEDULE					
4. PERFORMING ORGANIZATION REPORT NUMBER(S) HDL-TR-2150			5. MONITORING ORGANIZATION REPORT NUMBER(S)		
6a. NAME OF PERFORMING ORGANIZATION Harry Diamond Laboratories		6b. OFFICE SYMBOL (if applicable) SLCHD-ST-AP		7a. NAME OF MONITORING ORGANIZATION	
6c. ADDRESS (City, State, and ZIP Code) 2800 Powder Mill Road Adelphi, MD 20783-1197			7b. ADDRESS (City, State, and ZIP Code)		
8a. NAME OF FUNDING/SPONSORING ORGANIZATION Center for Night Vision and Electro-Optics		8b. OFFICE SYMBOL (if applicable)		9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER	
8c. ADDRESS (City, State, and ZIP Code) Ft. Belvoir, VA 22060			10. SOURCE OF FUNDING NUMBERS		
			PROGRAM ELEMENT NO. 612120.H25	PROJECT NO. AH25	TASK NO. WORK UNIT ACCESSION NO.
11. TITLE (Include Security Classification) Analysis of the Optical Spectra of Triply Ionized Transition Metal Ions in Yttrium Aluminum Garnet (YAG)					
12. PERSONAL AUTHOR(S) Clyde A. Morrison and Gregory A. Turner					
13a. TYPE OF REPORT Final		13b. TIME COVERED FROM 2/87 TO 10/87		14. DATE OF REPORT (Year, Month, Day) October 1988	
15. PAGE COUNT 42					
16. SUPPLEMENTARY NOTATION HDL project: 520851, AMS code: 612120.H250011					
17. COSATI CODES			18. SUBJECT TERMS (Continue on reverse if necessary and identify by block number)		
FIELD	GROUP	SUB-GROUP			
20	02		YAG, transition metals, optical spectra, $Ti^{3+}$ , $V^{3+}$ , $Cr^{3+}$ , $Mn^{3+}$ , $Fe^{3+}$ , $Co^{3+}$ , $Ni^{3+}$		
20	05		$Cu^{3+}$ , $Zn^{3+}$ , crystal field Yttrium, Aluminum, Garnet		
19. ABSTRACT (Continue on reverse if necessary and identify by block number)					
<p>An analysis of the reported spectra of five transition metal ions in yttrium aluminum garnet (YAG) is given. The reported data assumed that the ions occupied the <math>C_{3i}</math> site in YAG, and this was also assumed in the analysis. The point charge lattice sums, <math>A_{nm}</math>, were calculated for the <math>C_{3i}</math> site, and the concept of rotational invariance was used to obtain a set of crystal field parameters, <math>B_{nm}</math>, for the triply ionized <math>3d^N</math> electronic series. The energy levels for the entire (<math>Ti^{3+}</math>—<math>Zn^{3+}</math>) series of transition metal ions was calculated using the <math>B_{nm}</math> for the <math>C_{3i}</math> site. The spin-orbit interaction was included in the calculation.</p>					
20. DISTRIBUTION/AVAILABILITY OF ABSTRACT <input checked="" type="checkbox"/> UNCLASSIFIED/UNLIMITED <input type="checkbox"/> SAME AS RPT <input type="checkbox"/> DTIC USERS			21. ABSTRACT SECURITY CLASSIFICATION UNCLASSIFIED		
22a. NAME OF RESPONSIBLE INDIVIDUAL Clyde A. Morrison			22b. TELEPHONE (Include Area Code) (202) 394-2042		22c. OFFICE SYMBOL SLCHD-ST-AP

## Contents

	Page
1. Introduction .....	5
2. Free Ion and Crystal Field Interactions .....	6
3. Analysis in the Cubic Approximation for the $C_{3i}$ Site .....	7
4. Theoretical Crystal Field Parameters .....	20
4.1 Point Charge Crystal Field .....	20
4.2 Discussion of Results .....	33
References .....	36
Distribution .....	39

## Tables

1. Experimental Values of $F^{(2)}$ , $F^{(4)}$ , $\alpha$ , and $B_{40}$ for Triply Ionized Transition Metal Ions, $X^{3+}$ , in $Y_3Al_5O_{12}$ .....	7
2. Energy Levels of $Ti^{3+}$ in $Y_3Al_5O_{12}$ .....	8
3. Energy Levels of $V^{3+}$ in $Y_3Al_5O_{12}$ .....	8
4 - 8. Energy Levels of $Cr^{3+}$ in $Y_3Al_5O_{12}$ .....	9-13
9. Energy Levels of $Mn^{3+}$ in $Y_3Al_5O_{12}$ .....	14
10. Energy Levels of $Fe^{3+}$ in $Y_3Al_5O_{12}$ .....	15
11. Energy Levels of $Fe^{3+}$ in $Y_3Al_5O_{12}$ .....	17
12. Energy Levels of $Co^{3+}$ in $Y_3Al_5O_{12}$ .....	18
13. Crystallographic Data on $Y_3Al_5O_{12}$ .....	22
14. Lattice Sum, $A_{nm}$ , for $Al_2$ Ion in $24(d)$ ( $S_4$ ) Site in $Y_3Al_5O_{12}$ .....	23

# Tables (cont'd)

15. Lattice Sum, $A_{nm}$ , for Al Ion in 16(a) ( $C_{3i}$ ) Site in $Y_3Al_5O_{12}$ .....	23
16. Linear Least-Squares Best Fit Values of $\rho_2$ and $\rho_4$ for Triply Ionized Transition Metal Ions in Octahedral ( $C_{3i}$ ) Site in YAG .....	23
17. Linear Least-Squares Best Fit $F^{(k)}$ and $B_{nm}$ for Triply Ionized Transition Metal Ions in Octahedral Site ( $C_{3i}$ ) in YAG .....	23
18. Racah Parameters, $B$ and $C$ ; Macfarlane's parameters, $Dq$ , $v$ , and $v'$ ; and Ballhausen's parameters, $Dq$ , $D\sigma$ , and $D\tau$ , for Triply Ionized Transition Metal Ions in Octahedral ( $C_{3i}$ ) Site in YAG. ....	23
19. Energy Levels of $Ti^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 122.24$ .....	24
20. Energy Levels of $V^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 164.80$ .....	24
21. Energy Levels of $Cr^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 220.00$ .....	25
22. Energy Levels of $Mn^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 289.44$ .....	26
23. Energy Levels of $Fe^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 370.40$ .....	28
24. Energy Levels of $Co^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 484.80$ .....	29
25. Energy Levels of $Ni^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 599.20$ .....	31
26. Energy Levels of $Cu^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 728.80$ .....	32
27. Energy Levels of $Zn^{3+}$ in $Y_3Al_5O_{12}$ Using Parameters of Table 17 with $\alpha = 30$ and $\zeta = 923.88$ .....	33

## 1. Introduction

We report here on the analysis of the spectra of triply ionized transition metal ions,  $X^{3+}$ , with the electronic configuration  $3d^N$  in the host lattice  $Y_3Al_5O_{12}$  (YAG). The analysis is restricted predominantly to ions assumed to occupy the octahedral site ( $C_{3i}$ ). Most of the experimental data on the  $3d^N$  electronic configuration are for  $N < 5$ , with a large number of results for the  $3d^3$  configuration,  $Cr^{3+}$ . The possibility that the  $X^{3+}$  ions may occupy sites other than the  $C_{3i}$  site (dodecahedral,  $D_2$ , or tetragonal,  $S_4$ ) further complicates the analysis [1].\* Further, much of the experimental data used in the analysis has been taken at room temperature, and any possibility of observing detailed electronic line structure is lost. Despite these shortcomings in the reported data, we analyze the existing data and formulate a method of calculating a set of parameters which can be used as first approximates to experimental energy levels. These parameters are used to calculate the energy levels of the entire  $3d^N$  triply ionized transition elements. This latter calculation includes the full  $C_{3i}$  symmetry as well as spin-orbit coupling.



Assignment for	
1. [illegible]	<input checked="" type="checkbox"/>
2. [illegible]	<input type="checkbox"/>
3. [illegible]	<input type="checkbox"/>
4. [illegible]	
5. [illegible]	
6. [illegible]	
7. [illegible]	
8. [illegible]	
9. [illegible]	
10. [illegible]	
A-1	

\*References are listed at the end of the text (p 36).

## 2. Free Ion and Crystal Field Interactions

The interaction considered in the analysis given here has been described in detail previously [2] and will be covered only briefly here. The free ion terms in the interaction contain the Slater parameters  $F^{(2)}$  and  $F^{(4)}$ , and the configuration interaction contains the parameters  $\alpha$  and  $\gamma$ . In the analysis here we shall assume  $\gamma = 0$  throughout. The crystal field of  $C_{3i}$  symmetry for the  $nd^N$  configuration is

$$H_{CEF} = B_{20} \sum_{i=1}^N C_{20}(i) + B_{40} \sum_{i=1}^N C_{40}(i) + B_{43} \sum_{i=1}^N [C_{43}(i) - C_{4-3}(i)] , \quad (1)$$

$$C_{nm}(i) = \sqrt{4\pi/(2n+1)} Y_{nm}(\theta_i, \Phi_i) ,$$

where the crystal field parameter,  $B_{43}$ , is real and, for definiteness, with no loss in generality, we assume  $B_{43}$  positive. For the cubic approximation in equation (1), we have

$$B_{20} = 0 ,$$

$$B_{43} = \sqrt{10/7} |B_{40}| . \quad (2)$$

(The  $z$  axis of the crystal field is parallel to the  $\langle 111 \rangle$  cubic axis.) The relationship of the crystal field parameters  $B_{nm}$  to other notations is given by König and Kremer [3]. For ions in the  $S_4$  site, the crystal field interaction is taken as

$$H_{CEF} = B_{20} \sum_{i=1}^N C_{20}(i) + B_{40} \sum_{i=1}^N C_{40}(i) + B_{44} \sum_{i=1}^N [C_{44}(i) + C_{4-4}(i)] , \quad (3)$$

where the crystal field parameter  $B_{44}$  is real and assumed positive. In  $S_4$  symmetry, the total crystal field interaction contains terms with  $B_{32}$  and  $B_{52}$ . Both parameters are generally complex, but since we are considering interactions only in the  $3d^N$  configuration, we shall ignore these terms. Further, it should be noted that since the odd terms  $B_{32}$  and  $B_{52}$  are present, electric dipole transitions can exist, and if they are present, their absorption intensity should be much larger [1] than for ions in the  $C_{3i}$  site for which only magnetic dipole or electric quadrupole transitions are allowed. This, of course, has been recognized by others and has been used to identify the site occupancy of the ions, particularly  $\text{Fe}^{3+}$  in YAG. The cubic approximation (tetrahedral) is obtained by letting

$$B_{20} = 0$$

$$B_{44} = \sqrt{5/14} |B_{40}| . \quad (4)$$

in equation (3) [3].

### 3. Analysis in the Cubic Approximation for the $C_{3i}$ Site

The reported parameters for the various ions  $X^{3+}$  in YAG were used in a least-squares fit to the experimental data. Most of the data were analyzed in terms of  $Dq$ ,  $B$ , and  $C$ . The relation of these parameters to our parameters is

$$F^{(2)} = 7(7B + C)$$

$$F^{(4)} = 63C/5$$

$$B_{40} = -14 Dq ,$$

and  $B_{43}$  given by equation (2). This latter relation is for  $C_{3i}$  symmetry; however, for  $S_4$  symmetry sites,  $B_{40} = -21Dq$  and  $B_{44}$  is given by equation (4). Starting with the parameters given in the references in table 1 [4-13] and the experimental data from these references, we obtained a best least-squares fit to the experimental data by varying  $F^{(2)}$ ,  $F^{(4)}$ ,  $\alpha$ , and  $B_{40}$ . These results are given in table 1. Tables 2 through 12 give the experimental and calculated energy levels from our fitting. In these tables  $F^{(2)}$ ,  $F^{(4)}$ ,  $\alpha$ , and  $B_{40}$  were, in general, varied to obtain the best fit to the experimental data reported in the given references. For several ions the number of experimental energy levels did not exceed the number of parameters, and in these cases the results should be viewed accordingly. In particular the results obtained for  $Co^{3+}$  are very questionable because of the large rms value with only six experimental energy levels.

Table 1. Experimental values of  $F^{(2)}$ ,  $F^{(4)}$ ,  $\alpha$ , and  $B_{40}$  for triply ionized transition metal ions,  $X^{3+}$ , in  $Y_3Al_5O_{12}$

Ion	$nd^N$	$F^{(2)}$	$F^{(4)}$	$\alpha$	$B_{40}^{(a)}$	Ref No.	Table No.
Ti	3d <sup>1</sup>	—	—	—	-25928	4	2
V	3d <sup>2</sup>	52164	38344	70	-24104	5	3
Cr	3d <sup>3</sup>	60360	40020	0	-22442	6	4
Cr	3d <sup>3</sup>	54870	36073	97	-23116	7	5
Cr	3d <sup>3</sup>	56509	34680	110	-22941	8	6
Cr	3d <sup>3</sup>	58125	44929	0	-23978	9	7
Cr	3d <sup>3</sup>	54834	35666	122	-22960	10	8
Mn	3d <sup>4</sup>	61762	41454	0	-29419	11	9
Fe	3d <sup>5</sup>	49147	36324	23.8	-17580	12	10
Fe	3d <sup>5</sup>	49280	37297	0	-17703	13	11
Co	3d <sup>6</sup>	41033	22381	26.8	-23292	11	12

$$^a B_{43} = \sqrt{10/7} |B_{40}|$$

**Table 2. Energy levels**  
( $\text{cm}^{-1}$ ) of  $\text{Ti}^{3+}$  in  
 $\text{Y}_3\text{Al}_5\text{O}_{12}$

*[Theoretical levels  
were calculated with  
parameters in  
table 1]*

Level No.	IR <sup>a</sup>	Energy <sup>b</sup> (exp)	Energy (theor)	Free-ion state (%)
1	${}^2\text{T}_2$	0	-1	100 2D
2	${}^2\text{E}$	18519	18519	100 2D

<sup>a</sup>In this and other tables which follow, IR is the irreducible representation of the cubic group.

<sup>b</sup>Experimental energy levels are taken from reference 4.

**Table 3. Energy levels**  
( $\text{cm}^{-1}$ ) of  $\text{V}^{3+}$  in  
 $\text{Y}_3\text{Al}_5\text{O}_{12}$

*[Theoretical levels  
were calculated with  
parameters in  
table 1]*

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	${}^3\text{T}_1$	0	0	91 3F + 9 3P
2	${}^1\text{T}_2$	10256	10256	58 1D + 42 1G
3	${}^1\text{E}$	0	10468	51 1D + 49 1G
4	${}^3\text{T}_2$	16000	16000	100 3F
5	${}^1\text{A}_1$	0	22142	68 1G + 32 1S
6	${}^3\text{T}_1$	23530	23530	91 3P + 9 3F
7	${}^1\text{T}_2$	0	27546	58 1G + 42 1D
8	${}^1\text{T}_1$	0	30204	100 1G
9	${}^3\text{A}_2$	0	33217	100 3F
10	${}^1\text{E}$	0	44552	51 1G + 49 1D
11	${}^1\text{A}_1$	0	58381	68 1S + 32 1G

<sup>a</sup>Experimental energy levels are taken from reference 5.

Table 4. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Cr}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^4\text{A}_2$	0	124	100 4F
2	$^2\text{E}$	14500	15264	63 2G + 19 2H + 17 2D2
3	$^2\text{T}_1$	0	15994	41 2G + 35 2H + 20 2P
4	$^4\text{T}_2$	16700	16154	100 4F
5	$^2\text{T}_2$	0	22621	35 2H + 27 2D1 + 1 2D2
6	$^4\text{T}_1$	23200	23511	58 4F + 42 4P
7	$^2\text{A}_1$	0	28795	100 2G
8	$^2\text{T}_2$	31200	31187	58 2G + 38 2H + 3 2D2
9	$^2\text{T}_1$	31200	31725	54 2H + 25 2G + 21 2P
10	$^2\text{E}$	0	33498	49 2H + 40 2D2 + 10 2D1
11	$^4\text{T}_1$	36300	36498	58 4P + 42 4F
12	$^2\text{T}_1$	37800	36921	37 2H + 35 2P + 25 2F
13	$^2\text{T}_2$	43100	42673	56 2F + 41 2D2 + 2 2G
14	$^2\text{A}_2$	0	44357	100 2F
15	$^2\text{T}_1$	0	47324	68 2H + 32 2G
16	$^2\text{T}_2$	0	49268	26 2G + 25 2F + 22 2H
17	$^2\text{E}$	0	52404	43 2D2 + 24 2G + 17 2D1
18	$^2\text{T}_1$	55000	55274	71 2F + 23 2P + 5 2H
19	$^2\text{T}_2$	0	71313	61 2D1 + 22 2D2 + 13 2F
20	$^2\text{E}$	0	71538	71 2D1 + 16 2H + 13 2G

<sup>a</sup>Experimental energy levels are taken from reference 6.



Table 5. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Cr}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^4\text{A}_2$	0	138	100 4F
2	$^2\text{E}$	14559	14592	62 2G + 20 2D2 + 17 2H
3	$^2\text{T}_1$	15265	15247	38 2G + 33 2H + 24 2P
4	$^4\text{T}_2$	16700	16650	100 4F
5	$^2\text{T}_2$	21281	21278	30 2H + 30 2D1 + 20 2D2
6	$^4\text{T}_1$	23200	23146	51 4P + 49 4F
7	$^2\text{A}_1$	0	28858	100 2G
8	$^2\text{T}_2$	0	31407	57 2G + 40 2H + 3 2D2
9	$^2\text{T}_1$	0	31551	39 2H + 30 2G + 29 2P
10	$^2\text{E}$	0	32924	49 2H + 41 2D2 + 10 2D1
11	$^2\text{T}_1$	0	36171	52 2H + 25 2F + 22 2P
12	$^4\text{T}_1$	36300	36357	51 4F + 49 4P
13	$^2\text{T}_2$	0	40606	60 2F + 38 2D2 + 2 2G
14	$^2\text{A}_2$	0	42298	100 2F
15	$^2\text{T}_1$	0	48295	69 2H + 31 2G
16	$^2\text{T}_2$	0	48888	29 2G + 25 2H + 19 2F
17	$^2\text{E}$	0	50425	39 2D2 + 27 2D1 + 21 2G
18	$^2\text{T}_1$	0	53834	69 2F + 24 2P + 7 2H
19	$^2\text{T}_2$	0	68067	59 2D1 + 22 2D2 + 15 2F
20	$^2\text{E}$	0	70007	61 2D1 + 21 2H + 17 2G

<sup>a</sup>Experimental energy levels are taken from reference 7.

**Table 6. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Cr}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$**

*[Theoretical levels were calculated with parameters in table 1]*

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^4\text{A}_2$	0	16	100 4F
2	$^2\text{E}$	14616	14523	62 2G + 20 2D2 + 16 2H
3	$^2\text{T}_1$	15200	15255	39 2G + 32 2H + 25 2P
4	$^4\text{T}_2$	16400	16402	100 4F
5	$^2\text{R}_2$	21100	21110	31 2D1 + 29 2H + 21 2D2
6	$^4\text{T}_1$	23200	23187	52 4F + 48 4P
7	$^2\text{A}_1$	0	28579	100 2G
8	$^2\text{T}_2$	0	31307	57 2G + 40 2H + 3 2D2
9	$^2\text{T}_1$	0	31432	37 2H + 31 2P + 30 2G
10	$^2\text{E}$	0	32783	49 2H + 40 2D2 + 11 2D1
11	$^2\text{T}_1$	0	36300	54 2H + 24 2F + 22 2P
12	$^4\text{T}_1$	36300	36304	52 4P + 48 4F
13	$^2\text{T}_2$	0	41060	58 2F + 39 2D2 + 2 2G
14	$^2\text{A}_2$	0	42900	100 2F
15	$^2\text{T}_1$	0	48153	70 2H + 30 2G
16	$^2\text{T}_2$	0	48699	27 2G + 26 2H + 21 2F
17	$^2\text{E}$	0	50858	40 2D2 + 25 2D1 + 21 2G
18	$^2\text{T}_1$	0	54084	70 2F + 23 2P + 7 2H
19	$^2\text{T}_2$	0	68579	58 2D1 + 23 2D2 + 15 2F
20	$^2\text{E}$	0	69692	62 2D1 + 21 2H + 16 2G

<sup>a</sup>Experimental energy levels are taken from reference 8.

Table 7. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Cr}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^4\text{A}_2$	0	-132	100 4F
2	$^2\text{E}$	0	15593	61 2G + 19 2H + 18 2D2
3	$^2\text{T}_1$	0	16164	39 2G + 35 2H + 21 2P
4	$^4\text{T}_2$	16950	16995	100 4F
5	$^2\text{T}_2$	0	23329	38 2H + 27 2D1 + 17 2D2
6	$^4\text{T}_1$	23730	23783	50 4P + 50 4F
7	$^2\text{A}_1$	0	30400	100 2G
8	$^2\text{T}_2$	0	32556	58 2G + 37 2H + 4 2D2
9	$^2\text{T}_1$	0	32946	55 2H + 25 2G + 20 2P
10	$^2\text{E}$	0	34794	49 2H + 43 2D2 + 9 2D1
11	$^4\text{T}_1$	37540	37486	50 4F + 50 4P
12	$^2\text{T}_1$	0	37700	36 2H + 32 2P + 29 2F
13	$^2\text{T}_2$	0	42524	62 2F + 37 2D2 + 1 2G
14	$^2\text{A}_2$	0	43936	100 2F
15	$^2\text{T}_1$	0	49675	67 2H + 33 2G
16	$^2\text{T}_2$	0	51438	29 2G + 22 2D2 + 21 2H
17	$^2\text{E}$	0	52976	39 2D2 + 24 2D1 + 23 2G
18	$^2\text{T}_1$	0	56329	66 2F + 27 2P + 7 2H
19	$^2\text{T}_2$	0	72119	63 2D1 + 19 2D2 + 14 2F
20	$^2\text{E}$	0	74668	66 2D1 + 18 2H + 15 2G

<sup>a</sup>Experimental energy levels are taken from reference 9.

**Table 8. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Cr}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$**

*[Theoretical levels were calculated with parameters in table 1]*

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^4\text{A}_2$	0	1	100 4F
2	$^2\text{E}$	14542	14541	61 2G + 21 2D2 + 16 2H
3	$^2\text{T}_1$	15200	15199	38 2G + 32 2H + 25 2P
4	$^4\text{T}_2$	16400	16401	100 4F
5	$^2\text{T}_2$	21100	21100	31 2D1 + 29 2H + 21 2D2
6	$^4\text{T}_1$	0	22789	51 4P + 49 4F
7	$^2\text{A}_1$	0	28727	100 2G
8	$^2\text{T}_2$	0	31373	56 2G + 40 2H + 3 2D2
9	$^2\text{T}_1$	0	31385	35 2H + 32 2P + 31 2G
10	$^2\text{E}$	0	32741	49 2H + 41 2D2 + 10 2D1
11	$^2\text{T}_1$	0	35913	51 4F + 49 4P
12	$^4\text{T}_1$	0	36059	56 2H + 24 2F + 19 2P
13	$^2\text{T}_2$	0	40290	60 2F + 38 2D2 + 2 2G
14	$^2\text{A}_2$	0	42045	100 2F
15	$^2\text{T}_1$	0	48240	69 2H + 30 2G
16	$^2\text{T}_2$	0	48595	29 2G + 26 2H + 19 2F
17	$^2\text{E}$	0	50105	38 2D2 + 28 2D1 + 21 2G
18	$^2\text{T}_1$	0	53436	69 2F + 23 2P + 8 2H
19	$^2\text{T}_2$	0	67566	58 2D1 + 22 2D2 + 15 2F
20	$^2\text{E}$	0	69492	60 2D1 + 22 2H + 17 2G

<sup>a</sup>Experimental energy levels are taken from reference 10.

Table 9. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Mn}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^3T_1$	0	0	40 3H + 17 3G + 16 3F1
2	$^5E$	0	1087	100 5D
3	$^1T_2$	0	9356	35 1I + 28 1G2 + 23 1G1
4	$^1E$	0	10500	57 1I + 22 1D2 + 11 1G1
5	$^3E$	0	18226	84 3H + 15 3G + 1 3D
6	$^1A_1$	0	18869	38 1G1 + 24 1I + 20 1S2
7	$^3T_1$	0	19072	45 3H + 23 3G + 19 3P2
8	$^3T_2$	0	19913	41 3F2 + 32 3H + 13 3F1
9	$^3A_1$	0	21361	100 3G
10	$^5T_2$	22100	22100	100 5D
11	$^1T_2$	0	22154	94 3F2 + 6 3F1
12	$^3E$	0	24733	56 3D + 41 3G + 3 3H
13	$^1A_2$	0	27162	91 1I + 9 1F
14	$^1T_2$	27200	27200	61 1I + 23 1G2 + 7 1G1
15	$^1T_1$	0	29443	40 1I + 24 1G2 + 20 1G1
16	$^1E$	31500	31500	41 1D2 + 33 1G2 + 20 1I
17	$^3T_1$	0	32447	35 3G + 27 3H + 25 3P1
18	$^3T_2$	0	34073	39 3D + 32 3F1 + 23 3G
19	$^3T_1$	0	39507	52 3H + 32 3F2 + 6 3F1
20	$^3T_2$	0	40957	56 3H + 21 3F2 + 11 3D
21	$^1A_1$	0	41615	62 1G2 + 18 1S2 + 16 1G1
22	$^1T_1$	0	43564	37 1F + 24 1G1 + 22 1G2
23	$^3T_2$	0	43717	44 3G + 43 3D + 12 3H
24	$^1T_2$	0	44831	62 1F + 19 1D1 + 13 1G1
25	$^3T_2$	0	45135	40 3G + 33 3F2 + 24 3D
26	$^1T_2$	0	47185	85 1I + 9 1G2 + 3 1G1
27	$^3T_1$	0	48161	57 3P2 + 20 3G + 17 3F2
28	$^1E$	0	48262	43 1G2 + 26 1G1 + 22 1I
29	$^1A_1$	0	49786	41 1I + 29 1S2 + 18 1G2
30	$^1T_2$	0	56126	70 1D2 + 11 1I + 9 1G1
31	$^1A_2$	0	56846	91 1F + 9 1I
32	$^1T_1$	0	57683	53 1G2 + 46 1F + 1 1I
33	$^3T_1$	0	59751	71 3F1 + 19 3H + 6 3P2

**Table 9. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Mn}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  (cont'd)**

*[Theoretical levels were calculated with parameters in table 1]*

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
34	$^3\text{A}_2$	0	62517	94 3F1 + 6 3F2
35	$^1\text{E}$	0	67079	62 1G1 + 23 1G2 + 11 1D2
36	$^3\text{T}_1$	0	68610	51 3P1 + 27 3F2 + 13 3H
37	$^1\text{T}_1$	0	72164	55 1G1 + 43 1I + 1 1F
38	$^3\text{T}_2$	0	73296	53 3F1 + 19 3G + 19 3D
39	$^1\text{T}_2$	0	76810	28 1G1 + 28 1D1 + 22 1D2
40	$^1\text{A}_1$	0	84593	27 1S2 + 26 1I + 23 1G1
41	$^1\text{E}$	0	86023	79 1D1 + 20 1D2 + 1 1I
42	$^1\text{T}_2$	0	86877	41 1D1 + 23 1F + 17 1G2
43	$^1\text{A}_1$	0	108559	66 1S1 + 23 1G1 + 6 1S2

<sup>a</sup>Experimental energy levels are taken from reference 11.

**Table 10. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Fe}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$**

*[Theoretical levels were calculated with parameters in table 1]*

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^6\text{A}_1$	0	-2	100 6S
2	$^4\text{T}_1$	10400	10400	59 4G + 38 4P + 3 4F
3	$^2\text{T}_2$	0	10827	30 2I + 26 2H + 19 2F1
4	$^4\text{T}_2$	14300	14309	46 4G + 29 4F + 25 4D
5	$^2\text{A}_2$	0	20610	54 2I + 42 2F1 + 4 2F2
6	$^4(\text{A}_1, \text{E})$	20717	20800	100 4G
7	$^2\text{T}_1$	0	20931	58 2I + 24 2H + 14 2F1
8	$^2\text{T}_2$	0	22075	48 2I + 24 2F1 + 12 2H
9	$^4\text{T}_2$	22880	22679	55 4D + 43 4G + 2 4F
10	$^2\text{E}$	0	23473	33 2I + 31 2H + 15 2D3
11	$^4\text{E}$	24440	24605	100 4D
12	$^2\text{T}_1$	0	26897	32 2G2 + 21 2F1 + 20 2H
13	$^2\text{T}_2$	0	28889	30 2G2 + 25 2F2 + 14 2I
14	$^2\text{A}_1$	0	29251	62 2I + 19 2G2 + 13 2G1
15	$^4\text{T}_1$	0	30530	46 4F + 42 4P + 13 4G

Table 10. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Fe}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  (cont'd)

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
16	$^2\text{E}$	0	30665	47 2I + 26 2D3 + 13 2D1
17	$^2\text{T}_2$	0	32011	51 2I + 36 2D3 + 8 2D1
18	$^2\text{A}_1$	0	32131	44 2G2 + 37 2I + 11 2G1
19	$^4\text{A}_2$	0	33470	100 4F
20	$^2\text{E}$	0	34902	74 2G2 + 11 2I + 5 2D3
21	$^2\text{T}_2$	0	36135	45 2F2 + 26 2G2 + 16 2I
22	$^2\text{T}_1$	0	36592	68 2H + 30 2F1 + 1 2G2
23	$^2\text{A}_2$	0	36949	71 2F2 + 22 2I + 7 2F1
24	$^2\text{T}_1$	0	37093	48 2F2 + 39 2G2 + 9 2I
25	$^4\text{T}_1$	0	37704	51 4F + 29 4G + 20 4P
26	$^2\text{A}_1$	0	39945	69 2S + 27 2G2 + 4 2G1
27	$^2\text{E}$	0	40975	41 2H + 33 2D3 + 25 2D2
28	$^4\text{T}_2$	0	41888	69 4F + 21 4D + 10 4G
29	$^2\text{T}_1$	0	42283	37 2H + 19 2F2 + 18 2F1
30	$^2\text{T}_2$	0	43413	31 2F1 + 19 2H + 15 2G2
31	$^2\text{T}_1$	0	46706	43 2H + 30 2I + 16 2F1
32	$^2\text{A}_2$	0	46747	51 2F1 + 25 2F2 + 24 2I
33	$^2\text{E}$	0	49269	72 2G1 + 9 2H + 8 2D1
34	$^2\text{T}_2$	0	51523	53 2D2 + 13 2G1 + 12 2I
35	$^2\text{E}$	0	52402	54 2D2 + 11 2G1 + 11 2G2
36	$^2\text{T}_2$	0	53835	25 2D3 + 25 2H + 16 2I
37	$^2\text{T}_1$	0	54302	78 2G1 + 8 2P + 6 2F2
38	$^2\text{T}_2$	0	57447	47 2G1 + 13 2G2 + 12 2F2
39	$^2\text{A}_1$	0	59526	72 2G1 + 17 2S + 11 2G2
40	$^2\text{T}_1$	0	63651	86 2P + 5 2F2 + 5 2G2
41	$^2\text{T}_2$	0	73086	66 2D1 + 7 2H + 7 2D3
42	$^2\text{E}$	0	73251	67 2D1 + 13 2D3 + 12 2G1

<sup>a</sup>Experimental energy levels are taken from reference 12.

Table 11. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Fe}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	${}^6\text{A}_1$	0	-1	100 6S
2	${}^4\text{T}_1$	10360	10377	60 4G + 37 4P + 4 4F
3	${}^2\text{T}_2$	0	10680	32 2I + 27 2H + 18 2F1
4	${}^4\text{T}_2$	14200	14195	48 4G + 29 4F + 24 4D
5	${}^2\text{A}_2$	0	20438	56 2I + 41 2F1 + 4 2F2
6	${}^2\text{T}_1$	0	20610	60 2I + 24 2H + 13 2F1
7	${}^4(\text{A}_1, \text{E})$	20530	20627	100 4G
8	${}^2\text{T}_2$	0	21874	49 2I + 24 2F1 + 11 2H
9	${}^4\text{T}_2$	22940	22659	55 4D + 42 4G + 3 4F
10	${}^2\text{E}$	0	23329	36 2I + 30 2H + 14 2D3
11	${}^4\text{E}$	24450	24707	100 4D
12	${}^2\text{T}_1$	0	26902	33 2G2 + 21 2F1 + 20 2H
13	${}^2\text{T}_2$	0	28814	30 2G2 + 24 2F2 + 17 2I
14	${}^2\text{A}_1$	0	28912	68 2I + 16 2G2 + 11 2G1
15	${}^2\text{E}$	0	30639	47 2I + 25 2D3 + 13 2D1
16	${}^4\text{T}_1$	0	30741	47 4F + 42 4P + 11 4G
17	${}^2\text{T}_1$	0	31947	48 2G2 + 31 2I + 13 2G1
18	${}^2\text{T}_2$	0	31985	48 2I + 37 2D3 + 8 2D1
19	${}^4\text{A}_2$	0	33542	100 4F
20	${}^2\text{E}$	0	34885	74 2G2 + 9 2I + 6 2D3
21	${}^2\text{T}_2$	0	36158	45 2F2 + 25 2G2 + 15 2I
22	${}^2\text{T}_1$	0	36678	69 2H + 30 2F1 + 1 2G2
23	${}^2\text{A}_2$	0	36961	71 2F2 + 21 2I + 8 2F1
24	${}^2\text{T}_1$	0	37090	48 2F2 + 39 2G2 + 9 2I
25	${}^4\text{T}_1$	0	37850	50 4F + 29 4G + 22 4P
26	${}^2\text{A}_1$	0	40168	69 2S + 26 2G2 + 4 2G1
27	${}^2\text{E}$	0	41121	40 2H + 35 2D3 + 25 2D2
28	${}^4\text{T}_2$	0	42022	69 4F + 21 4D + 10 4G
29	${}^2\text{T}_1$	0	42379	36 2H + 20 2F2 + 18 2F1
30	${}^2\text{T}_2$	0	43560	31 2F1 + 20 2H + 15 2G2
31	${}^2\text{T}_1$	0	46693	42 2H + 29 2I + 17 2F1
32	${}^2\text{A}_2$	0	46953	52 2F1 + 25 2F2 + 23 2I
33	${}^2\text{E}$	0	49272	73 2G1 + 9 2H + 8 2D1



Table 11. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Fe}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  (cont'd)

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
34	$^2T_2$	0	51725	50 2D2 + 16 2G1 + 13 2I
35	$^2E$	0	52668	56 2D2 + 11 2G2 + 10 2H
36	$^2T_2$	0	54029	26 2D3 + 23 2H + 16 2F1
37	$^2T_1$	0	54335	78 2G1 + 8 2P + 7 2F2
38	$^2T_2$	0	57575	47 2G1 + 13 2G2 + 12 2F2
39	$^2A_1$	0	59643	71 2G1 + 17 2S + 10 2G2
40	$^2T_1$	0	63922	86 2P + 5 2F2 + 5 2G2
41	$^2T_2$	0	73556	67 2D1 + 7 2H + 7 2D3
42	$^2E$	0	73691	68 2D1 + 13 2D3 + 12 2G1

<sup>a</sup>Experimental energy levels are taken from reference 13.

Table 12. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Co}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
1	$^1A_1$	0	-232	35 1G1 + 33 1I + 20 1G2
2	$^3T_1$	0	11110	41 3H + 25 3F2 + 20 3P1
3	$^3T_2$	14900	14919	38 3G + 24 3H + 23 3F1
4	$^1T_1$	15600	15464	55 1I + 28 1G1 + 18 1G2
5	$^5T_2$	0	16390	100 5D
6	$^1T_2$	22400	22672	27 1I + 21 1D2 + 20 1F
7	$^3T_1$	26600	26802	64 3H + 26 3F2 + 7 3F1
8	$^3T_2$	0	27829	53 3H + 42 3G + 5 3F2
9	$^3E$	29700	29359	52 3G + 29 3H + 19 3D
10	$^3T_1$	0	29359	48 3G + 32 3P2 + 11 3H
11	$^1E$	0	31597	45 1G2 + 30 1G1 + 22 1I
12	$^3T_2$	0	31673	54 3D + 43 3F2 + 4 3G
13	$^1T_2$	0	32124	69 1I + 12 1G2 + 10 1D2
14	$^1A_1$	0	32352	52 1I + 28 1S2 + 10 1S1
15	$^5E$	0	33027	100 5D
16	$^3A_2$	0	35959	74 3F1 + 26 3F2

Table 12. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Co}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  (cont'd)

[Theoretical levels were calculated with parameters in table 1]

Level No.	IR	Energy <sup>a</sup> (exp)	Energy (theor)	Free-ion state (%)
17	$^1\text{T}_1$	0	36366	39 1G2 + 32 1F + 25 1I
18	$^1\text{A}_2$	0	37076	58 1F + 42 1I
19	$^1\text{T}_2$	0	37610	27 1D2 + 27 1F + 23 1I
20	$^3\text{T}_1$	0	40727	32 3H + 29 3F2 + 17 3F1
21	$^1\text{E}$	0	41617	38 1D2 + 36 1G1 + 19 1D1
22	$^3\text{E}$	0	43870	69 3H + 29 3G + 2 3D
23	$^3\text{T}_1$	0	44577	33 3G + 29 3P2 + 22 3H
24	$^3\text{T}_2$	0	44791	31 3F2 + 23 3H + 19 3D
25	$^1\text{T}_2$	0	44988	43 1G1 + 30 1I + 15 1D2
26	$^3\text{A}_1$	0	45760	100 3G
27	$^3\text{E}$	0	48475	79 3D + 19 3G + 2 3H
28	$^1\text{E}$	0	48642	48 1I + 37 1G2 + 9 1D1
29	$^1\text{A}_1$	0	50412	57 1G2 + 21 1S2 + 18 1S1
30	$^1\text{A}_2$	0	50540	58 1I + 42 1F
31	$^1\text{T}_2$	0	51406	43 1G2 + 22 1I + 18 1F
32	$^1\text{T}_2$	0	51995	74 3F2 + 26 3F1
33	$^1\text{T}_1$	0	52204	61 1F + 17 1I + 15 1G1
34	$^3\text{T}_1$	0	57039	57 3P1 + 19 3F2 + 12 3P2
35	$^3\text{T}_2$	0	57173	61 3F1 + 18 3F2 + 15 3D
36	$^1\text{T}_2$	0	60211	40 1G1 + 25 1I + 15 1D2
37	$^1\text{A}_1$	0	60860	37 1S2 + 37 1G1 + 10 1S1
38	$^1\text{E}$	0	61751	54 1D1 + 34 1D2 + 7 1G2
39	$^1\text{T}_1$	0	62434	54 1G1 + 36 1G2 + 7 1F
40	$^3\text{T}_1$	0	64466	53 3F1 + 19 3H + 18 3G
41	$^1\text{E}$	0	71113	30 1G1 + 23 1D2 + 20 1I
42	$^1\text{T}_2$	0	76690	59 1D1 + 19 1F + 10 1D2
43	$^1\text{A}_1$	0	87780	57 1S1 + 26 1G1 + 8 1S2

<sup>a</sup>Experimental energy levels are taken from reference 11.

## 4. Theoretical Crystal Field Parameters

### 4.1 Point Charge Crystal Field

The crystallographic data of Euler and Bruce [14] on YAG are given in table 13. The yttrium site (24c) and the  $\text{Al}_2$  site (24d) have been chosen to have the principal axis of each site along the crystallographic  $c$ -axis. The crystal field components,  $A_{nm}$ , for the  $\text{Al}_2$  site are given in table 14. The various contributions to the  $A_{nm}$  are individually given, but most of our analysis will center about the point charge values of  $A_{nm}$  (for the dipole contribution, the polarizabilities of Schmidt et al [15] are used). In tetrahedral coordination ( $\text{Al}_2$ ), the lattice sum components should be such that  $|A_{44}/A_{40}| = (5/14)^{1/2}$ , which for the point charge given in table 14,  $|A_{44}/A_{40}| = 0.3928$ , which is considerably less than the true tetrahedral value. From these results one would expect that a tetrahedral approximation to the analysis of ions occupying the  $\text{Al}_2$  site would be a rather poor one. The corresponding  $A_{nm}$  for the  $\text{Al}_1$  site (16a) are given in table 15. In this table the crystal field components were first computed in the crystallographic coordinates and then rotated so that the principal axis of  $C_{3i}$  was parallel to the (111) crystallographic axis. If the site were pure octahedral, we would have  $|A_{43}/A_{40}| = (10/7)^{1/2} = 1.1952$ , and for the point charge values of  $A_{40}$  and  $A_{43}$  of table 15,  $|A_{43}/A_{40}| = 1.1243$ , which is quite close to the octahedral value. Of course the  $A_{20}$  value is significantly different from the octahedral value of zero.

In the point charge or point multipole theory of the crystal field interaction, the theoretical crystal field parameters are given by

$$B_{nm} = \rho_n A_{nm} \quad , \quad (6)$$

where the  $A_{nm}$  are the crystal field components given in tables 14 and 15 and  $\rho_n$  is an effective radial expectation value. To relate the theoretical  $B_{nm}$  with those obtained by fitting the experimental data, we use the concept of a rotational invariant [16,17], which we define as

$$S_n(B) = \left( \sum_{m=-n}^n B_{nm}^* B_{nm} \right)^{1/2} \quad . \quad (7)$$

For  $n = 4$  and  $C_{3i}$  symmetry, we have

$$S_4(B) = (B_{40}^2 + 2B_{43}^2)^{1/2} , \quad (8)$$

and from equation (6) we obtain

$$S_4(B) = \rho_4 S_4(A) , \quad (9)$$

where

$$S_4(A) = (A_{40}^2 + 2A_{43}^2)^{1/2} .$$

For cubic symmetry and using equation (2), we obtain

$$S_4(B) = \sqrt{27/7} |B_{40}^e| , \quad (10)$$

where the  $B_{40}^e$  are the experimental values given in table 1. The results of table 15 for the point charge  $A_{4m}$  along with the  $B_{40}^e$  given in table 1 were used to obtain  $\rho_4$  with

$$\rho_4 = \frac{\sqrt{27/7} |B_{40}^e|}{S_4(A)} . \quad (11)$$

In ions for which we have more than one set of data, we selectively average the  $\rho_4$  obtained from equation (11). If we assume that the wavefunctions expand from the Hartree-Fock value, then we assume that

$$\rho_n = \langle r^n \rangle_{HF} \tau^n , \quad (12)$$

where this result is obtained by assuming the radial wavefunction  $R(r) = R_{HF}(\tau r)$ . Using equation (12) we have

$$\frac{1}{\tau^2} = (\rho_4 / \langle r^4 \rangle_{HF})^{1/2} , \quad (13)$$

and from equation (12) for  $n = 2$ , we obtain

$$\rho_2 \approx \langle r^2 \rangle_{HF} (\rho_4 / \langle r^4 \rangle_{HF})^{1/2} , \quad (14)$$

which can be used to obtain theoretical values of  $B_{2m}$  by equation (6). The  $\rho_2$  and  $\rho_4$  determined by equations (14) and (11) were used in equation (6) to determine values of the crystal field parameter for  $\text{Cr}^{3+}$ . These values of  $B_{nm}$  along with the  $F^{(k)}$  of table 8 and with  $\zeta = 220 \text{ cm}^{-1}$  (80 percent of the free ion value) were used to calculate the ground-state splitting ( $0.53 \text{ cm}^{-1}$ ) and the splitting of the  ${}^2E$  level ( $19.6 \text{ cm}^{-1}$ ). Both of these calculated splittings were much too large. Consequently, the value of  $B_{20}$  was varied to obtain a best fit to these splittings, and a value of  $B_{20} = 2832 \text{ cm}^{-1}$  was found, which is approximately 0.5301 times the value given by equation (14). Thus all the values of  $\rho_2$  given by equation (14) were multiplied by 0.5301. The values of  $\rho_4$  were obtained from equation (11) and these values are given in table 16 for the triply ionized  $3d^N$  transition metal ion series. In obtaining these results, we used the Hartree-Fock values of  $\langle r^n \rangle$  given by Fraga et al [18]. Further, the results obtained from the experimental values of  $\rho_4$  were linear least-squares fit to obtain values of  $\rho_2$  and  $\rho_4$  for ions not reported.

The values of  $\rho_2$  and  $\rho_4$  given in table 16 were used in equation (6) to obtain crystal field parameters  $B_{nm}$  which were used (fixed) to fit the experimental data given in tables 2 through 12 with  $F^{(2)}$  and  $F^{(4)}$  varying. The resulting  $F^{(2)}$  and  $F^{(4)}$  were linear least-squares fit, and the final  $F^{(k)}$  and  $B_{nm}$  are given in table 17. Since most of the experimental data are analyzed in terms of Racah parameters  $B$  and  $C$ , as well as various crystal field parameters, two sets of these parameters (Macfarlane's [19] and Bellhauser's [20]) are given in table 18.

The set of parameters given in table 17, along with the spin-orbit coupling set at 80 percent of the free ion value [21] and  $\alpha = 30 \text{ cm}^{-1}$ , was used to calculate the energy levels for the  $3d^1$  through the  $3d^9$  ions. The results are given in tables 19 through 27.

Table 13. Crystallographic data on  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Cubic  $\text{O}_h^{10}$  ( $Ia3d$ ) No. 230,  $z = 8$ ]

Ion	Site	Symm	$x^a$	$y$	$z$	$q$	$\alpha(\text{\AA}^3)^{(b)}$
$\text{Al}_1$	16(a)	$\text{C}_{3i}$	0	0	0	3	0.0530
$\text{Al}_2$	24(d)	$\text{S}_4$	0	1/4	3/8	3	0.0530
Y	24(c)	$\text{D}_2$	0	1/4	1/8	3	0.870
O	96(h)	$\text{C}_1$	-0.0306	-0.0512	0.1500	-2	1.349

<sup>a</sup>X-ray data,  $a = 12.000 \text{ \AA}$ , reference 14.

<sup>b</sup>Taken from reference 15.

Table 14. Lattice sum,  $A_{nm}$  ( $\text{cm}^{-1}/\text{\AA}^n$ ), for  $\text{Al}_2$  ion in 24(d) ( $S_4$ ) site in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

$A_{nm}$	Point charge	Self-induced	Dipole	Total
$A_{20}$	6,355	-2,604	14,013	17,765
$\text{Re}A_{32}$	-27,522	8,609	-11,957	-30,870
$\text{Im}A_{32}$	37,839	-11,913	6,332	32,258
$A_{40}$	-25,089	11,879	-8,516	-21,726
$\text{Re}A_{44}$	-3,763	1,614	1,964	-185.1
$\text{Im}A_{44}$	-9,108	4,740	-2,875	-7,243
$\text{Re}A_{52}$	-2,931	2,287	-3,498	-4,142
$\text{Im}A_{52}$	4,328	-3,207	3,640	4,762
$ A_{44} $	9,855	—	—	7,245

Table 16. Linear least-squares best fit values of  $\rho_2$  and  $\rho_4$  for triply ionized transition metal ions in octahedral ( $C_{3v}$ ) site in YAG

Ion	$3d^N$	$\rho_2(\text{\AA}^2)$	$\rho_4(\text{\AA}^4)$
Ti	$d^1$	0.4301	1.3431
V	$d^2$	0.4234	1.3039
Cr	$d^3$	0.4142	1.2647
Mn	$d^4$	0.4050	1.2255
Fe	$d^5$	0.3963	1.1864
Co	$d^6$	0.3855	1.1472
Ni	$d^7$	0.3756	1.1080
Cu	$d^8$	0.3660	1.0689
Zn	$d^9$	0.3567	1.0297

Table 18. Racah parameters,  $B$  and  $C$ ; Macfarlane's parameters,  $Dq$ ,  $v$ , and  $v'$ ; and Ballhausen's parameters,  $Dq$ ,  $D\sigma$ , and  $D\tau$ , for triply ionized transition metal ions in octahedral ( $C_{3v}$ ) site in YAG.

Ion	$3d^N$	$B$	$C$	$Dq^{(a)}$	$v$	$v'$	$Dq^{(b)}$	$D\sigma$	$D\tau$
Ti	$d^1$	—	—	1839	758.0	-776.1	1810	-421.9	76.15
V	$d^2$	748.1	2868	1786	747.7	-759.0	1757	-413.5	73.93
Cr	$d^3$	783.0	3088	1732	735.5	-741.1	1704	-404.5	71.71
Mn	$d^4$	817.8	3309	1678	723.3	-723.3	1651	-395.5	69.49
Fe	$d^5$	852.7	3529	1625	712.7	-705.9	1599	-387.0	67.27
Co	$d^6$	887.6	3749	1571	695.8	-685.7	1546	-376.5	65.04
Ni	$d^7$	922.5	3969	1517	681.6	-666.8	1493	-366.8	62.82
Cu	$d^8$	957.3	4190	1464	668.3	-648.3	1441	-357.4	60.60
Zn	$d^9$	—	—	1410	655.8	-630.2	1387	-348.3	58.38

<sup>a</sup>Reference 19. <sup>b</sup>Reference 20.

Table 15. Lattice sum,  $A_{nm}$  ( $\text{cm}^{-1}/\text{\AA}^n$ ), for  $\text{Al}_1$  ion in 16(a) ( $C_{3v}$ ) site in  $\text{Y}_3\text{Al}_5\text{O}_{12}$

[Rotated so that the z-axis is parallel to the (111) crystallographic axis]

$A_{nm}$	Point charge	Self-induced	Dipole	Total
$A_{20}$	6,836	-1,107	-13,553	-7,823
$A_{40}$	-20,054	8,166	3,273	-8,615
$\text{Re}A_{43}$	2,813	-1,422	6,253	7,644
$\text{Im}A_{43}$	-22,370	8,639	2,348	-11,383
$ A_{43} $	22,546	—	—	13,711

Table 17. Linear least-squares best fit values of  $F^{(k)}$  ( $\text{cm}^{-1}$ ) and  $B_{nm}$  ( $\text{cm}^{-1}$ ) for triply ionized transition metal ions in the octahedral site ( $C_{3v}$ ) in YAG

Ion	$3d^N$	$F^{(2)}$	$F^{(4)}$	$B_{20}$	$B_{40}$	$B_{43}$
Ti	$d^1$	—	—	2953	-26934	30280
V	$d^2$	56732	36132	2895	-26148	29397
Cr	$d^3$	59983	38912	2832	-25363	28514
Mn	$d^4$	63234	41688	2769	-24577	27631
Fe	$d^5$	66485	44463	2709	-23792	26748
Co	$d^6$	69736	47239	2635	-23006	25865
Ni	$d^7$	72987	50015	2568	-22221	24982
Cu	$d^8$	76238	52790	2502	-21435	24099
Zn	$d^9$	—	—	2438	-20650	23216

**Table 19. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Ti}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 122.24$  ( $\text{cm}^{-1}$ )**

Level No.	IR <sup>a</sup>	Energy	Free-ion state (%)
1	6	0	100 2D
2	4,5	128	100 2D
3	4,5	868	100 2D
4	4,5	18773	100 2D
5	6	18787	100 2D

<sup>a</sup>4,5 and 6 are abbreviations for the  $\Gamma_{4,5}$  and  $\Gamma_6$  irreducible representations of the  $C_{3i}$  double group. Levels labeled 6 are also doublets.

**Table 20. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{V}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 164.80$  ( $\text{cm}^{-1}$ )**

Level No.	IR <sup>a</sup>	Energy	Free-ion state (%)
1	1	0	90 3F + 10 3P
2	2,3	13	90 3F + 10 3P
3	2,3	156	93 3F + 7 3P
4	2,3	313	92 3F + 8 3P
5	1	386	93 3F + 7 3P
6	1	441	92 3F + 8 3P
7	2,3	10582	54 1D + 46 1G
8	1	11086	62 1D + 37 1G
9	2,3	11181	54 1D + 46 1G
10	1	16405	100 3F
11	2,3	16413	100 3F
12	1	16734	100 3F
13	1	16737	100 3F
14	2,3	16783	100 3F
15	2,3	16826	100 3F
16	1	23377	68 1G + 31 1S
17	1	25453	88 3P + 12 3F
18	2,3	25456	88 3P + 12 3F
19	1	26436	92 3P + 8 3F
20	1	26447	92 3P + 7 3F
21	2,3	26504	93 3P + 7 3F
22	2,3	26565	93 3P + 7 3F
23	2,3	28764	57 1G + 43 1D
24	1	28805	62 1G + 37 1D
25	1	31367	100 1G
26	2,3	31837	99 1G + 1 1D
27	1	34650	99 3F + 1 3P
28	2,3	34651	99 3F + 1 3P
29	1	46539	51 1G + 49 1D
30	2,3	61136	69 1S + 31 1G

<sup>a</sup>Labels 1 and 2,3 are abbreviations for the  $\Gamma_1$  and  $\Gamma_{2,3}$  irreducible representations of the  $C_{3i}$  group.

Table 21. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Cr}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and

$\zeta = 220.00$  ( $\text{cm}^{-1}$ )

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
1	4,5	0	100 4F	31	4,5	34715	50 2H + 38 2D2 + 10 2D1
2	6	0	100 4F	32	4,5	37935	29 2H + 20 4F + 20 4P
3	4,5	15099	61 2G + 19 2H + 17 2D2	33	6	38024	28 4P + 27 4F + 19 2H
4	6	15119	61 2G + 19 2H + 17 2D2	34	4,5	38036	23 2H + 22 4F + 22 4P
5	6	15788	41 2G + 34 2H + 20 2P	35	4,5	38130	52 4P + 48 4F
6	4,5	15816	40 2G + 34 2H + 21 2P	36	4,5	38273	32 4P + 32 4F + 14 2H
7	4,5	16062	38 2G + 32 2H + 23 2P	37	6	38311	24 4P + 22 4F + 21 2H
8	4,5	17163	100 4F	38	4,5	38381	27 4P + 26 4F + 18 2H
9	4,5	17205	99 4F + 1 2G	39	4,5	39898	58 4P + 41 4F
10	6	17250	99 4F + 1 2G	40	6	39916	58 4P + 41 4F + 1 2P
11	4,5	17307	98 4F + 1 2G	41	4,5	43545	56 2F + 40 2D2 + 2 2G
12	6	17630	99 4F + 1 2G	42	6	43586	56 2F + 40 2D2 + 2 2G
13	4,5	17637	99 4F + 1 2G	43	4,5	43768	63 2F + 35 2D2 + 1 2G
14	4,5	22114	30 2H + 29 2D1 + 20 2D2	44	4,5	45378	99 2F
15	6	22750	35 2H + 28 2D1 + 18 2D2	45	4,5	49752	65 2H + 33 2G + 2 2D2
16	4,5	22846	35 2H + 28 2D1 + 18 2D2	46	6	49843	64 2H + 33 2G + 1 2D2
17	4,5	24405	58 4F + 41 4P	47	4,5	50494	69 2H + 29 2G + 1 2P
18	6	24408	58 4F + 41 4P	48	4,5	51132	30 2G + 22 2H + 19 2F
19	4,5	24999	52 4F + 47 4P	49	4,5	51728	27 2H + 26 2G + 21 2F
20	6	25012	51 4F + 48 4P	50	6	51817	26 2G + 26 2H + 22 2F
21	4,5	25025	51 4F + 49 4P	51	4,5	54078	41 2D2 + 22 2D1 + 22 2G
22	4,5	25036	50 4F + 50 4P	52	6	54090	41 2D2 + 22 2D1 + 22 2G
23	4,5	29991	100 2G	53	4,5	57404	69 2F + 23 2P + 7 2H
24	4,5	32345	57 2G + 37 2H + 4 2D2	54	6	57410	68 2F + 24 2P + 7 2H
25	6	32378	58 2G + 36 2H + 4 2D2	55	4,5	57542	69 2F + 24 2P + 6 2H
26	4,5	32662	46 2H + 30 2G + 23 2P	56	6	72614	60 2D1 + 19 2D2 + 14 2F
27	4,5	32823	52 2G + 44 2H + 3 2D2	57	4,5	72734	60 2D1 + 19 2D2 + 14 2F
28	4,5	32975	50 2H + 28 2G + 19 2P	58	4,5	73980	60 2D1 + 23 2D2 + 13 2F
29	6	33077	50 2H + 25 2G + 21 2P	59	6	74606	64 2D1 + 17 2H + 14 2G
30	6	34709	49 2H + 38 2D2 + 10 2D1	60	4,5	74642	64 2D1 + 17 2H + 14 2G

<sup>a</sup>Labels 4,5 and 6 are abbreviations for the  $\Gamma_{4,5}$  and  $\Gamma_6$  irreducible representations of the  $C_{3i}$  double group. Levels labeled 6 are also doublets.



Table 22. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Mn}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 289.44$  ( $\text{cm}^{-1}$ )

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
1	1	0	99 5D	44	1	21137	95 3G + 3 3F2 + 1 3H
2	2,3	12	99 5D	45	1	21140	33 1G1 + 23 1I + 23 1S2
3	1	24	99 5D	46	1	21370	91 3F2 + 7 3F1 + 1 5D
4	2,3	35	99 5D	47	2,3	21376	90 3F2 + 7 3F1 + 1 3G
5	2,3	50	100 5D	48	1	24200	52 3D + 44 3G + 4 3H
6	1	60	100 5D	49	2,3	24201	52 3D + 44 3G + 4 3H
7	1	66	100 5D	50	1	24202	52 3D + 44 3G + 4 3H
8	1	3295	38 3H + 18 3G + 15 3F1	51	2,3	24204	52 3D + 44 3G + 4 3H
9	1	3354	39 3H + 18 3G + 15 3F1	52	2,3	27088	60 1I + 23 1G2 + 8 1G1
10	2,3	3455	39 3H + 17 3G + 15 3F1	53	1	27310	64 1I + 21 1G2 + 7 1G1
11	2,3	3614	41 3H + 17 3G + 15 3F1	54	1	27328	92 1I + 8 1F
12	2,3	3918	41 3H + 15 3G + 14 3F1	55	1	28890	40 1I + 24 1G2 + 20 1G1
13	1	3921	41 3H + 15 3G + 14 3F1	56	2,3	29238	39 1I + 26 1G2 + 20 1G1
14	1	12948	32 1I + 30 1G2 + 23 1G1	57	2,3	30403	39 3G + 30 3H + 19 3P1
15	2,3	13108	35 1I + 28 1G2 + 23 1G1	58	2,3	30440	39 3G + 29 3H + 19 3P1
16	2,3	14297	59 1I + 20 1D2 + 10 1G1	59	1	30467	40 3G + 29 3H + 18 3P1
17	1	16316	98 5D + 1 3H	60	1	30477	41 3G + 28 3H + 18 3P1
18	2,3	16324	98 5D + 1 3H	61	2,3	30797	43 3G + 27 3H + 19 3P1
19	2,3	16347	99 5D	62	1	30799	44 3G + 28 3H + 19 3P1
20	1	16936	97 5D + 1 3H + 1 3P2	63	2,3	31190	41 1D2 + 32 1G2 + 16 1I
21	1	16936	97 5D + 1 3H + 1 3P2	64	1	32618	33 3D + 26 3G + 24 3F1
22	2,3	17031	97 5D + 1 3H + 1 3F2	65	1	32645	32 3D + 26 3G + 24 3F1
23	2,3	17114	97 5D + 1 3H + 1 3F2	66	2,3	32656	31 3D + 27 3G + 23 3F1
24	1	17173	97 5D + 1 3F2 + 1 3H	67	2,3	32682	29 3D + 28 3G + 23 3F1
25	1	17205	97 5D + 1 3F2 + 1 3H	68	2,3	32819	33 3G + 31 3D + 24 3F1
26	2,3	17260	97 5D + 1 3F2 + 1 3H	69	1	32821	33 3G + 31 3D + 24 3F1
27	2,3	18031	84 3H + 14 3G + 1 3D	70	1	34849	48 3H + 32 3F2 + 6 3G
28	2,3	18049	84 3H + 14 3G + 1 3D	71	1	34873	48 3H + 33 3F2 + 5 3G
29	1	18051	84 3H + 14 3G + 1 3D	72	2,3	34905	47 3H + 33 3F2 + 6 3F1
30	1	18061	84 3H + 14 3G + 1 3D	73	2,3	34944	47 3H + 33 3F2 + 6 3F1
31	1	18344	46 3H + 20 3G + 17 3P2	74	1	35176	51 3H + 30 3F2 + 6 3F1
32	2,3	18402	45 3H + 22 3G + 17 3P2	75	2,3	35182	50 3H + 29 3F2 + 6 3F1
33	1	18640	46 3H + 19 3P2 + 18 3G	76	1	36672	49 3H + 25 3F2 + 18 3D
34	1	18742	45 3H + 21 3G + 19 3P2	77	2,3	36677	49 3H + 26 3F2 + 18 3D
35	2,3	18799	45 3H + 20 3G + 19 3P2	78	1	37256	46 3H + 28 3D + 16 3F2
36	2,3	18879	44 3H + 20 3G + 19 3P2	79	1	37264	45 3H + 28 3D + 16 3F2
37	2,3	19273	38 3F2 + 32 3H + 13 3F1	80	2,3	37274	45 3H + 29 3D + 17 3F2
38	2,3	19333	40 3F2 + 31 3H + 13 3F1	81	2,3	37297	44 3H + 30 3D + 17 3F2
39	1	19379	40 3F2 + 32 3H + 13 3F1	82	2,3	39307	45 3D + 41 3G + 11 3H
40	1	19424	41 3F2 + 32 3H + 13 3F1	83	1	39310	46 3D + 41 3G + 11 3H
41	2,3	19650	40 3F2 + 32 3H + 12 3F1	84	2,3	39313	46 3D + 41 3G + 11 3H
42	1	19662	41 3F2 + 32 3H + 12 3F1	85	1	39316	46 3D + 41 3G + 11 3H
43	2,3	21132	93 3G + 4 3F2 + 1 3H	86	1	40104	47 3G + 34 3F2 + 15 3D

Table 22. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Mn}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 289.44$  ( $\text{cm}^{-1}$ ) (cont'd)

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
87	1	40146	46 3G + 34 3F2 + 15 3D	115	1	54639	68 3F1 + 18 3H + 9 3P1
88	2,3	40226	46 3G + 33 3F2 + 16 3D	116	2,3	55026	72 3F1 + 18 3H + 5 3P1
89	2,3	40378	46 3G + 33 3F2 + 17 3D	117	1	55055	72 3F1 + 17 3H + 5 3P1
90	2,3	40681	44 3G + 27 3D + 26 3F2	118	1	58711	91 3F1 + 8 3F2
91	1	40701	44 3G + 28 3D + 26 3F2	119	2,3	58711	91 3F1 + 8 3F2
92	1	41720	38 1G2 + 23 1S2 + 19 1I	120	1	61623	50 3P1 + 22 3F2 + 16 3P2
93	2,3	42473	41 1F + 23 1I + 17 1G1	121	2,3	61627	50 3P1 + 22 3F2 + 16 3P2
94	1	42813	83 1I + 6 1G2 + 3 1D1	122	2,3	62141	44 3P1 + 24 3F2 + 12 3F1
95	2,3	43437	78 1I + 8 1G2 + 5 1F	123	2,3	62260	44 3P1 + 23 3F2 + 12 3P2
96	1	43446	40 1F + 19 1I + 18 1G2	124	1	62342	44 3P1 + 23 3F2 + 12 3P2
97	1	43537	51 3P2 + 22 3F2 + 12 3G	125	1	62385	44 3P1 + 22 3F2 + 13 3P2
98	1	43565	49 3P2 + 20 3F2 + 12 3G	126	2,3	63350	62 1G1 + 24 1G2 + 8 1D2
99	2,3	43603	46 3P2 + 19 3F2 + 11 3G	127	1	64278	63 1G1 + 36 1I + 1 1F
100	2,3	43706	45 3P2 + 17 3F2 + 11 3G	128	1	64891	59 3F1 + 16 3D + 15 3G
101	2,3	43867	22 1F + 19 3P2 + 14 1G2	129	2,3	64892	58 3F1 + 16 3D + 14 3G
102	2,3	43905	40 1F + 11 1G1 + 10 3P2	130	2,3	65357	60 1G1 + 34 1I + 2 1F
103	1	43983	49 1F + 12 3P2 + 10 1G1	131	2,3	66284	54 3F1 + 15 3D + 15 3G
104	1	44088	42 3P2 + 20 3F2 + 13 1F	132	2,3	66292	54 3F1 + 15 3D + 14 3G
105	2,3	44116	40 3P2 + 19 3F2 + 10 1G2	133	1	66299	53 3F1 + 16 3D + 14 3G
106	1	46037	41 1G2 + 27 1I + 18 1S2	134	1	66306	54 3F1 + 16 3D + 14 3G
107	2,3	51578	61 1D2 + 10 1I + 9 1G1	135	2,3	72473	37 1G1 + 27 1G2 + 19 1D2
108	1	51653	71 1D2 + 9 1I + 9 1G1	136	1	73177	37 1G1 + 24 1G2 + 15 1D2
109	1	52486	81 1F + 11 1G2 + 7 1I	137	1	75774	34 1G1 + 24 1I + 18 1S2
110	2,3	53409	49 1G2 + 38 1F + 6 1D2	138	1	78667	54 1D1 + 18 1F + 12 1G1
111	1	53533	52 1F + 45 1G2 + 3 1I	139	2,3	81044	57 1D1 + 18 1F + 10 1G2
112	2,3	54472	68 3F1 + 18 3H + 10 3P1	140	2,3	82543	77 1D1 + 20 1D2 + 1 1G2
113	2,3	54537	68 3F1 + 18 3H + 9 3P1	141	1	102454	73 1S1 + 13 1G1 + 11 1S2
114	1	54606	68 3F1 + 18 3H + 9 3P1				

<sup>a</sup>Labels 1 and 2,3 are abbreviations for the  $\Gamma_1$  and  $\Gamma_{2,3}$  irreducible representations of the  $C_{3i}$  group.

Table 23. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Fe}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 370.40$  ( $\text{cm}^{-1}$ )

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
1	4,5	0.00	100 6S	44	6	34446	30 2G2 + 24 2F1 + 21 2H
2	6	0.187	100 6S	45	4,5	34478	29 2G2 + 23 2F1 + 21 2H
3	4,5	0.262	100 6S	46	4,5	34972	30 2G2 + 23 2F1 + 20 2H
4	4,5	12196	21 4G + 18 2I + 16 2H	47	4,5	37052	35 2I + 23 2G2 + 14 2F2
5	4,5	12725	38 4G + 26 4P + 9 2I	48	4,5	37848	49 2I + 20 2G2 + 10 2F2
6	6	12729	38 4G + 26 4P + 10 2I	49	6	37898	29 2G2 + 23 2F2 + 19 2I
7	6	13046	57 4G + 38 4P + 4 4F	50	4,5	37950	30 2I + 28 2G2 + 17 2F2
8	4,5	13105	57 4G + 39 4P + 3 4F	51	6	38553	28 4P + 24 4F + 19 2I
9	4,5	13156	57 4G + 39 4P + 3 4F	52	4,5	38634	33 4P + 28 4F + 14 2I
10	4,5	13480	37 4G + 25 4P + 11 2I	53	4,5	38949	30 2I + 18 4P + 16 4F
11	6	13608	20 2I + 20 4G + 18 2H	54	6	39088	25 2I + 22 4P + 19 4F
12	4,5	13618	20 4G + 20 2I + 17 2H	55	4,5	39657	39 4P + 38 4F + 14 4G
13	4,5	18323	50 4G + 28 4F + 21 4D	56	6	39802	37 4P + 36 4F + 14 4G
14	6	18336	50 4G + 28 4F + 21 4D	57	4,5	39848	39 4P + 39 4F + 15 4G
15	4,5	18861	47 4G + 28 4F + 25 4D	58	4,5	39900	38 4P + 37 4F + 14 4G
16	6	18871	48 4G + 28 4F + 24 4D	59	4,5	40763	47 2I + 35 2D3 + 9 2D1
17	4,5	18872	49 4G + 28 4F + 23 4D	60	6	40814	45 2I + 35 2D3 + 9 2D1
18	4,5	18884	49 4G + 28 4F + 23 4D	61	4,5	40869	41 2I + 34 2D3 + 9 2D1
19	4,5	25784	52 2I + 39 2F1 + 3 2F2	62	4,5	41825	52 2G2 + 22 2I + 14 2G1
20	4,5	25958	46 2I + 18 2H + 17 4G	63	4,5	43883	99 4F
21	6	25990	45 2I + 22 4G + 19 2H	64	6	43887	99 4F
22	4,5	26314	40 2I + 29 4G + 17 2H	65	6	45110	68 2G2 + 10 2I + 5 2H
23	4,5	26633	89 4G + 5 2I + 3 2F1	66	4,5	45176	70 2G2 + 10 2I + 5 2D3
24	4,5	26723	95 4G + 2 2I + 1 2F1	67	4,5	46241	60 2H + 28 2F1 + 5 2G2
25	6	26737	95 4G + 2 2I + 1 2F1	68	4,5	46325	55 2H + 22 2F1 + 11 2F2
26	4,5	26956	78 4G + 13 2I + 5 2H	69	6	46373	42 2H + 18 2F2 + 17 2F1
27	4,5	26982	73 4G + 17 2I + 6 2H	70	6	46735	26 2F2 + 22 2H + 18 2G2
28	6	26986	78 4G + 14 2I + 5 2H	71	4,5	46883	34 2F2 + 20 2G2 + 14 2H
29	4,5	27891	46 2I + 23 2F1 + 11 2H	72	4,5	47324	48 2F2 + 22 2G2 + 15 2I
30	6	28000	45 2I + 23 2F1 + 12 2H	73	4,5	47617	51 2F2 + 20 2I + 19 2G2
31	4,5	28003	43 2I + 21 2F1 + 12 2H	74	4,5	48403	33 2F2 + 28 2G2 + 16 4F
32	4,5	29247	43 4D + 35 4G + 6 2H	75	4,5	48445	40 4F + 19 4G + 13 4P
33	6	29289	43 4D + 34 4G + 7 2H	76	6	48533	40 2F2 + 37 2G2 + 7 2I
34	4,5	29418	52 4D + 41 4G + 3 4F	77	4,5	48542	37 2F2 + 19 4F + 14 2G2
35	4,5	29554	56 4D + 41 4G + 3 4F	78	4,5	48565	34 4F + 19 2F2 + 17 4G
36	6	29686	40 4D + 27 4G + 10 2I	79	6	48674	49 4F + 24 4G + 16 4P
37	4,5	29829	38 4D + 25 4G + 11 2I	80	4,5	48813	53 4F + 26 4G + 17 4P
38	4,5	29912	33 4D + 22 4G + 14 2I	81	6	49682	56 4F + 27 4G + 15 4P
39	6	29935	29 4D + 20 4G + 16 2I	82	4,5	49730	55 4F + 26 4G + 15 4P
40	4,5	32302	99 4D	83	4,5	52620	69 2S + 24 2G2 + 5 2G1
41	6	32303	99 4D	84	6	53040	42 2H + 30 2D3 + 22 2D2
42	4,5	32307	99 4D	85	4,5	53069	42 2H + 30 2D3 + 22 2D2
43	4,5	32313	99 4D	86	4,5	54048	31 2H + 22 2F2 + 18 2F1

Table 23. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Fe}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 370.40$  ( $\text{cm}^{-1}$ ) (cont'd)

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
87	4,5	54476	65 4F + 19 4D + 10 4G	107	4,5	68429	51 2D2 + 16 2G1 + 9 2G2
88	4,5	54514	63 4F + 19 4D + 9 4G	108	6	68481	47 2D2 + 16 2G1 + 11 2H
89	6	54575	61 4F + 18 4D + 9 4G	109	4,5	69067	31 2H + 22 2D3 + 18 2I
90	4,5	54731	49 4F + 16 4D + 10 2H	110	6	69666	21 2D3 + 20 2H + 16 2D2
91	6	54923	59 4F + 18 4D + 9 4G	111	4,5	69869	22 2D3 + 21 2H + 15 2F1
92	4,5	54981	65 4F + 20 4D + 10 4G	112	6	71446	77 2G1 + 6 2F2 + 5 2P
93	6	55138	29 2H + 21 2F2 + 14 2F1	113	4,5	71597	76 2G1 + 6 2F2 + 5 2P
94	4,5	55154	25 2H + 20 4F + 18 2F2	114	4,5	72031	79 2G1 + 8 2F2 + 7 2P
95	4,5	55831	28 2F1 + 19 2H + 17 2G2	115	4,5	75303	44 2G1 + 13 2D1 + 13 2G2
96	6	55938	27 2F1 + 18 2G2 + 17 2H	116	6	75729	53 2G1 + 13 2F2 + 12 2G2
97	4,5	56215	29 2F1 + 18 2H + 17 2G2	117	4,5	75826	52 2G1 + 13 2G2 + 11 2F2
98	4,5	59435	41 2H + 30 2I + 17 2F1	118	4,5	78268	72 2G1 + 17 2S + 10 2G2
99	6	59620	42 2H + 30 2I + 16 2F1	119	4,5	84683	89 2P + 4 2G2 + 4 2F2
100	4,5	59912	41 2F1 + 25 2I + 24 2F2	120	4,5	84860	86 2P + 4 2F2 + 4 2G2
101	4,5	60505	34 2H + 25 2I + 20 2F1	121	6	84892	87 2P + 4 2F2 + 4 2G2
102	4,5	64728	60 2G1 + 10 2H + 9 2D1	122	6	95434	67 2D1 + 10 2D3 + 6 2H
103	6	64792	60 2G1 + 10 2H + 9 2D1	123	4,5	95438	67 2D1 + 10 2D3 + 6 2H
104	4,5	66844	62 2D2 + 12 2G1 + 9 2F2	124	6	96691	65 2D1 + 15 2D3 + 12 2G1
105	6	67257	51 2D2 + 22 2G1 + 10 2I	125	4,5	96724	65 2D1 + 15 2D3 + 11 2G1
106	4,5	67316	51 2D2 + 22 2G1 + 10 2I	126	4,5	97034	63 2D1 + 10 2D3 + 9 2G1

<sup>a</sup>Labels 4,5 and 6 are abbreviations for the  $\Gamma_{4,5}$  and  $\Gamma_6$  irreducible representations of the  $C_{3i}$  double group. Levels labeled 6 are also doublets.

Table 24. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Co}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 484.80$  ( $\text{cm}^{-1}$ )

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
1	2,3	0	100 5D	14	1	6537	45 3H + 23 3F2 + 16 3P1
2	1	8	100 5D	15	1	6825	44 3H + 25 3F2 + 15 3P1
3	1	165	100 5D	16	2,3	6940	46 3H + 24 3F2 + 14 3P1
4	2,3	250	100 5D	17	1	7272	44 3H + 24 3F2 + 14 3P1
5	2,3	373	99 5D	18	1	11037	34 3H + 33 3G + 16 3F1
6	1	512	97 5D + 1 1I	19	2,3	11108	34 3H + 34 3G + 16 3F1
7	1	528	99 5D	20	2,3	11349	39 3G + 29 3H + 16 3F1
8	1	1023	87 5D + 4 1I + 3 1G1	21	2,3	11622	36 3G + 32 3H + 16 3F1
9	2,3	1057	99 5D	22	1	11741	36 3G + 32 3H + 16 3F1
10	2,3	1061	99 5D	23	1	11808	34 3H + 33 3G + 16 3F1
11	1	128	31 1I + 26 1G1 + 19 1G2	24	2,3	15231	57 1I + 22 1G1 + 18 1G2
12	2,3	6378	44 3H + 23 3F2 + 16 3P1	25	1	15438	53 1I + 23 1G1 + 20 1G2
13	2,3	6465	44 3H + 23 3F2 + 16 3P1	26	1	16256	99 5D

Table 24. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Co}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 484.80$  ( $\text{cm}^{-1}$ ) (cont'd)

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
27	1	16261	99 5D	71	2,3	36321	55 3H + 18 3G + 13 3F2
28	2,3	16273	99 5D	72	1	36445	58 3H + 15 3G + 14 3F2
29	2,3	16289	99 5D	73	2,3	36471	51 3H + 17 3G + 12 3F2
30	1	16296	99 5D	74	1	36563	52 3H + 16 3G + 9 3F2
31	1	16316	99 5D	75	2,3	36813	60 3H + 27 3G + 7 3F2
32	2,3	16317	99 5D	76	1	36823	57 3H + 21 3G + 13 3F2
33	1	20401	66 3H + 27 3F2 + 5 3F1	77	2,3	36872	59 3H + 32 3G + 2 3D
34	2,3	20404	65 3H + 26 3F2 + 5 3F1	78	1	36961	58 3H + 31 3G + 6 3F2
35	2,3	20643	65 3H + 25 3F2 + 6 3F1	79	1	37021	52 1I + 33 1F + 7 3H
36	2,3	20749	65 3H + 25 3F2 + 6 3F1	80	1	37050	46 3H + 23 3G + 8 1I
37	1	20828	67 3H + 25 3F2 + 6 3F1	81	2,3	37098	47 3H + 20 3G + 8 3F2
38	1	20854	66 3H + 25 3F2 + 6 3F1	82	2,3	38359	15 3D + 14 1F + 14 3F2
39	1	21777	48 3G + 47 3H + 4 3F2	83	1	38614	26 3D + 25 3F2 + 19 3H
40	1	21795	48 3G + 48 3H + 3 3F2	84	1	38801	17 1F + 17 3D + 13 1G2
41	2,3	21832	51 3H + 44 3G + 4 3F2	85	2,3	38868	16 1F + 14 3F2 + 14 1I
42	2,3	21898	53 3H + 41 3G + 5 3F2	86	1	39292	30 3D + 26 3F2 + 18 3H
43	2,3	22466	54 3G + 43 3H + 2 3F2	87	2,3	39475	30 3D + 26 3F2 + 18 3H
44	1	22469	54 3G + 43 3H + 1 3F2	88	1	39564	19 1F + 15 3D + 13 3F2
45	1	23628	26 1I + 21 3G + 16 3H	89	2,3	39653	31 3D + 26 3F2 + 18 3H
46	1	23832	45 3G + 30 3P2 + 13 3H	90	2,3	39784	97 3G + 1 3F2 + 1 3H
47	2,3	23863	44 3G + 33 3P2 + 11 3H	91	1	39787	96 3G + 1 3F2 + 1 3H
48	2,3	23877	40 3G + 27 3H + 12 3D	92	2,3	41737	29 3P2 + 26 3G + 23 3F2
49	1	24045	49 3G + 32 3H + 13 3D	93	2,3	41828	31 3P2 + 25 3G + 23 3F2
50	2,3	24059	48 3G + 28 3H + 12 3D	94	1	41919	32 3P2 + 25 3G + 22 3F2
51	2,3	24265	46 3G + 22 3P2 + 18 3H	95	1	41952	34 3P2 + 25 3G + 22 3F2
52	1	24362	40 3G + 21 3P2 + 10 1I	96	2,3	42600	33 3P2 + 28 3G + 20 3F2
53	2,3	24365	46 3G + 30 3P2 + 9 3H	97	1	42613	34 3P2 + 28 3G + 20 3F2
54	1	24504	49 3G + 27 3P2 + 12 3H	98	2,3	43976	33 1D2 + 23 1I + 23 1G1
55	1	24626	37 3G + 15 3H + 14 3P2	99	1	44046	80 3D + 15 3G + 2 1I
56	2,3	24957	37 1I + 14 1D2 + 11 1F	100	1	44061	82 3D + 15 3G + 2 3H
57	2,3	27783	49 3D + 47 3F2 + 1 3G	101	2,3	44067	81 3D + 15 3G + 2 3H
58	2,3	27830	49 3D + 47 3F2 + 1 3G	102	2,3	44097	82 3D + 15 3G + 2 3H
59	1	27868	49 3D + 47 3F2 + 2 3G	103	1	45367	50 1I + 19 1D2 + 19 1G1
60	1	27878	47 3D + 47 3F2 + 2 1I	104	2,3	45741	46 1I + 24 1D2 + 19 1G1
61	2,3	28094	53 3D + 43 3F2 + 2 1I	105	2,3	49708	41 1I + 31 1G2 + 18 1D2
62	1	28104	55 3D + 44 3F2 + 1 3F1	106	1	49839	60 1F + 38 1I + 1 1G1
63	2,3	30678	46 1G2 + 24 1G1 + 23 1I	107	1	50546	40 1S2 + 40 1G2 + 11 1S1
64	2,3	30883	48 1I + 17 1G2 + 16 1D2	108	2,3	52683	72 1F + 10 1I + 10 1G1
65	1	31302	48 1I + 15 1D2 + 12 1G2	109	1	53530	76 1F + 9 1I + 7 1G1
66	1	31751	50 1I + 18 1S2 + 13 1G2	110	1	54027	51 3F1 + 47 3F2 + 1 3P1
67	2,3	32911	50 3F2 + 46 3F1 + 2 1I	111	2,3	54032	51 3F1 + 47 3F2 + 1 3P1
68	1	32942	48 3F2 + 44 3F1 + 4 1I	112	1	55108	46 1G2 + 33 1F + 11 1G1
69	1	35653	27 1G2 + 24 1I + 17 3H	113	2,3	55641	48 1G2 + 34 1F + 8 1G1
70	2,3	35706	31 1G2 + 20 1I + 16 3H	114	1	57473	58 3P1 + 12 3F1 + 9 3H

**Table 24.** Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Co}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 484.80$  ( $\text{cm}^{-1}$ ) (cont'd)

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
115	1	57542	56 3P1 + 14 3F1 + 9 3H	129	1	64043	61 3F1 + 11 3P2 + 7 3F2
116	2,3	57570	55 3P1 + 15 3F1 + 9 3H	130	2,3	65366	58 3F1 + 15 3P2 + 8 3P1
117	2,3	57791	48 3P1 + 23 3F1 + 9 3H	131	1	65388	60 3F1 + 13 3P2 + 8 3G
118	2,3	58032	57 3P1 + 15 3F1 + 10 3H	132	2,3	65395	57 3F1 + 15 3P2 + 9 3P1
119	1	58061	57 3P1 + 14 3F1 + 10 3H	133	1	65399	60 3F1 + 14 3P2 + 8 3G
120	2,3	59977	63 3F1 + 15 3F2 + 7 3D	134	1	69443	62 1G1 + 33 1G2 + 3 1I
121	2,3	60007	64 3F1 + 15 3F2 + 8 3D	135	2,3	69469	59 1G1 + 32 1G2 + 4 1F
122	1	60048	66 3F1 + 16 3F2 + 9 3D	136	1	70869	53 1G1 + 22 1G2 + 14 1S2
123	1	60099	64 3F1 + 15 3F2 + 9 3D	137	2,3	73112	41 1D2 + 34 1G1 + 17 1G2
124	2,3	60776	68 3F1 + 19 3F2 + 9 3D	138	2,3	76771	78 1D1 + 10 1G1 + 6 1I
125	1	60806	68 3F1 + 19 3F2 + 9 3D	139	2,3	87513	73 1D1 + 15 1D2 + 9 1F
126	2,3	61798	66 1G1 + 20 1I + 9 1D2	140	1	88855	71 1D1 + 16 1D2 + 9 1F
127	1	62077	62 1G1 + 19 1I + 10 1D2	141	1	108376	71 1S1 + 13 1S2 + 11 1G1
128	2,3	63996	63 3F1 + 11 3P2 + 7 3F2				

<sup>a</sup>Labels 1 and 2,3 are abbreviations for the  $\Gamma_1$  and  $\Gamma_{2,3}$  irreducible representations of the  $C_{3i}$  group.

**Table 25.** Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Ni}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 599.20$  ( $\text{cm}^{-1}$ )

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
1	4,5	0	94 4F + 6 4P	22	4,5	22407	61 2P + 21 2H + 13 2F
2	6	354	92 4F + 7 4P	23	4,5	22948	62 2P + 22 2H + 13 2F
3	4,5	385	94 4F + 5 4P	24	4,5	25548	87 4P + 12 4F + 1 2P
4	4,5	908	90 4F + 6 4P + 2 2G	25	6	25648	85 4P + 12 4F + 1 2P
5	6	910	91 4F + 6 4P + 2 2G	26	4,5	26478	94 4P + 6 4F
6	4,5	1109	94 4F + 6 4P	27	6	26799	91 4P + 6 4F + 1 2H
7	4,5	4682	42 2G + 28 2H + 13 2D2	28	4,5	26989	92 4P + 5 4F + 1 2H
8	6	4729	43 2G + 29 2H + 13 2D2	29	4,5	27143	91 4P + 5 4F + 1 2P
9	6	14038	97 4F + 2 2G + 1 2D2	30	4,5	29650	88 4F + 5 4P + 3 2H
10	4,5	14066	98 4F + 1 2G	31	6	29656	87 4F + 5 4P + 4 2H
11	4,5	14305	98 4F + 1 2G	32	4,5	30250	97 2G + 1 2H + 1 2P
12	6	14350	97 4F + 2 2G	33	4,5	31055	55 2H + 28 2G + 9 2D1
13	4,5	14383	95 4F + 3 2G + 1 2D2	34	6	31307	54 2H + 24 2G + 9 2D1
14	4,5	14624	98 4F + 1 2G	35	4,5	31594	59 2H + 22 2G + 8 2D1
15	4,5	17383	50 2G + 46 2H + 2 4F	36	6	33728	66 2H + 19 2G + 12 2P
16	6	17542	53 2G + 42 2H + 2 4F	37	4,5	33838	66 2H + 21 2G + 11 2P
17	4,5	17995	51 2G + 46 2H + 2 4F	38	4,5	33937	61 2H + 27 2G + 10 2P
18	4,5	18248	59 2G + 16 2H + 15 2D2	39	4,5	35857	53 2H + 37 2D2 + 8 2D1
19	6	18272	58 2G + 16 2H + 16 2D2	40	6	35909	52 2H + 37 2D2 + 9 2D1
20	4,5	18424	57 2G + 19 2D2 + 15 2H	41	4,5	38962	72 2D2 + 12 2G + 11 2H
21	6	22265	60 2P + 21 2H + 12 2F	42	4,5	39156	68 2D2 + 17 2H + 11 2G

**Table 25. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Ni}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 599.20$  ( $\text{cm}^{-1}$ ) (cont'd)**

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
43	6	39210	69 2D2 + 18 2H + 10 2G	52	4,5	48499	100 2F
44	4,5	40856	48 2H + 38 2F + 13 2G	53	6	51917	43 2F + 24 2P + 18 2H
45	4,5	41769	45 2F + 40 2H + 9 2G	54	4,5	52066	44 2F + 24 2P + 18 2H
46	6	41894	45 2F + 40 2H + 11 2G	55	4,5	52796	48 2F + 25 2P + 16 2H
47	4,5	42731	73 2F + 15 2H + 8 2D1	56	4,5	68497	65 2D1 + 31 2D2 + 3 2H
48	6	43482	77 2F + 11 2H + 8 2D1	57	6	68511	64 2D1 + 32 2D2 + 3 2H
49	4,5	43799	78 2F + 10 2D1 + 9 2H	58	4,5	71390	76 2D1 + 12 2F + 6 2H
50	6	46346	53 2G + 17 2H + 15 2D1	59	6	72396	75 2D1 + 12 2F + 7 2D2
51	4,5	46352	53 2G + 16 2H + 15 2D1	60	4,5	72890	73 2D1 + 13 2F + 8 2D2

<sup>a</sup>Labels 4,5 and 6 are abbreviations for the  $\Gamma_{4,5}$  and  $\Gamma_6$  irreducible representations of the  $C_{3i}$  double group. Levels labeled 6 are also doublets.

**Table 26. Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Cu}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 728.80$  ( $\text{cm}^{-1}$ )**

Level No.	IR <sup>a</sup>	Energy	Free ion state (%)	Level No.	IR <sup>a</sup>	Energy	Free ion state (%)
1	1	0	100 3F	16	1	26964	85 1G + 10 1S + 3 3F
2	2,3	3	100 3F	17	2,3	29617	74 1D + 21 1G + 3 3P
3	2,3	1400	281 3F + 13 1D + 6 1G	18	1	303' -	71 1D + 25 1G + 3 3P
4	2,3	14434	99 3F	19	2,3	34653	80 1G + 15 3P + 5 3F
5	1	14583	100 3F	20	1	34780	44 1G + 42 3P + 14 3F
6	1	14814	100 3F	21	2,3	35026	57 3P + 29 3F + 13 1G
7	2,3	15000	98 3F + 1 1D + 1 1G	22	1	35114	70 3P + 27 3F + 1 1D
8	1	15162	100 3F	23	2,3	35133	61 3P + 30 3F + 7 1G
9	2,3	15969	53 1D + 24 1G + 22 3F	24	1	35388	56 1G + 30 3P + 14 3F
10	1	21784	74 3F + 24 3P + 2 1G	25	2,3	36821	71 3P + 27 3F + 1 1G
11	2,3	22141	74 3F + 26 3P	26	1	36855	71 3P + 27 3F + 1 1G
12	1	22698	72 3F + 27 3P	27	1	46631	75 1G + 25 1D
13	1	22893	68 3F + 31 3P	28	2,3	46686	68 1G + 32 1D
14	2,3	23128	67 3F + 32 3P	29	2,3	47877	77 1G + 23 1D
15	2,3	23313	65 3F + 33 3P + 1 1G	30	1	72868	89 1S + 11 1G

<sup>a</sup>Labels 1 and 2,3 are abbreviations for the  $\Gamma_1$  and  $\Gamma_{2,3}$  irreducible representations of the  $C_{3i}$  group.

**Table 27.** Energy levels ( $\text{cm}^{-1}$ ) of  $\text{Zn}^{3+}$  in  $\text{Y}_3\text{Al}_5\text{O}_{12}$  using parameters of table 17 with  $\alpha = 30$  and  $\zeta = 923.88^a$  ( $\text{cm}^{-1}$ )

Level No.	IR <sup>b</sup>	Energy	Free ion state (%)
1	6	0	100 2D
2	4,5	105	100 2D
3	4,5	13303	100 2D
4	4,5	14616	100 2D
5	6	15116	100 2D

<sup>a</sup>Hartree-Fock value (ref 18).

<sup>b</sup>Labels 4,5 and 6 are abbreviations for the  $\Gamma_{4,5}$  and  $\Gamma_6$  irreducible representations of the  $C_{3i}$  double group. Levels labeled 6 are also doublets.

## 4.2 Discussion of Results

Because of the sparse experimental data on the fine structure, a detailed comparison of the calculation given in tables 19 through 27 is not possible. The only ion with reliable fine structure reported is  $\text{Cr}^{3+}$  [7, 9, 10, 22], and these data were used to determine the crystal field parameter  $B_{20}$  for  $\text{Cr}^{3+}$ .

The results given for  $\text{Ti}^{3+}$  in table 19 show considerable splitting of the lower levels, but since only room temperature absorption was reported by Karpov et al [4], no comparison can be made. The absorption spectra at room temperature were reported by Bantien et al [23] as having bands at 25,600, 19,800, and 17,100  $\text{cm}^{-1}$ , with the band at 25,600  $\text{cm}^{-1}$  being very strong compared to the other two bands. The two lower energy bands were assumed to be  $3d^1 \rightarrow 3d^1$  transitions, but the high-energy band is unknown and under investigation.

The energy levels of  $\text{V}^{3+}$  given in table 20 show considerable splitting of the ground state ( $^3T_1$  in cubic approximation) so that if the absorption spectra were taken at low temperature rather than at room temperature as reported by Weber and Riseberg [5], the lines should be much narrower. Surprisingly, the spin-orbit and the two-fold field  $B_{20}$  do not mix the states much, and the usual assumption of spin-forbidden transitions in the analysis of the absorption spectra should be reasonably valid. It should be interesting to analyze the spectra taken at low temperature.



There are a considerable number of papers reporting the experimental data and analysis of experimental data on  $\text{Cr}^{3+}$  in YAG. The calculated splitting of the ground state,  $0.62 \text{ cm}^{-1}$ , is slightly larger than the experimental value [10]  $0.53 \text{ cm}^{-1}$ , but the splitting of the  $R$  lines calculated to be  $19.5 \text{ cm}^{-1}$  compares well with the experimental value [10]  $19.6 \text{ cm}^{-1}$  (levels 3 and 4 in table 21). The calculated centroid of the  $R$  lines is shifted approximately  $600 \text{ cm}^{-1}$  from the experimental value reported by Henry et al [10]. Comparison of the results in table 21 with the experimental results given in tables 4 through 7 shows, in general, that the calculated values are slightly higher than the experimental results. Slight adjustment of the parameters can bring the calculated values into better agreement with any one set of the experimental results. Also, mixing of the states by the spin-orbit does not occur below level number 32 ( $37,935 \text{ cm}^{-1}$ ) in table 21, which should make identification of the fine structure somewhat easier.

Using the parameters of table 17, we calculated the energy levels for  $\text{Mn}^{3+}$ , and the results are given in table 22. For this set of parameters, the high spin state  ${}^5D$  is the lowest, which contradicts the assumptions made by Arsenov and Sviridov [11] in analyzing the experimental data. However, only three energy levels were reported so that a re-analysis of the experimental data would be inconclusive. We found that if we reduced  $F^{(4)}$  from  $41,688 \text{ cm}^{-1}$  to  $33,332 \text{ cm}^{-1}$ , the ground state became mixed with the  ${}^3H$  levels (the low spin ground states assumed in ref 11).

The energy levels of  $\text{Fe}^{3+}$  given in table 23 were obtained using the parameters given in table 17 with  $\alpha = 30 \text{ cm}^{-1}$  and  $\gamma = 0$ . The ground state ( ${}^6A_1$  in the cubic approximation) is split by the spin orbit and the two-fold field,  $B_{20}$ . We have been unable to find any reported measurement of this ground state splitting. The spin-orbit coupling mixes the spin of a large number of levels, and the calculated energy levels appear to be considerably higher than experiment. We have not tried to fit the experimental data starting with the parameters of table 23. A comparison of the parameters obtained here for  $\text{Fe}^{3+}$  in YAG with the parameters for  $\text{Fe}^{3+}$  in  $\text{Ho}_3\text{Al}_5\text{O}_{12}$  or  $\text{Er}_3\text{Al}_5\text{O}_{12}$  shows that the  $F^{(k)}$  for YAG are larger but  $B_{20}$  is smaller than these compounds [2]. The  $B_{40}$  and  $B_{43}$  crystal field parameters are approximately the same. The experimental data for  $\text{Fe}^{3+}$  in the rare earth garnets were considerably different than reported for  $\text{Fe}^{3+}$  in YAG [12,13].

The energy levels for  $\text{Co}^{3+}$  were calculated using the parameters given in table 17 with  $\alpha = 30 \text{ cm}^{-1}$  and  $\gamma = 0$ ; the results are given in table 24. As for  $\text{Mn}^{3+}$ , the calculated ground state disagrees with that given by Arsenev and Sviridov [11]. However, we have observed that a reduction of the parameter  $\alpha$  causes the ground state, which in table 24 is predominately  $^5D$ , to become significantly mixed with  $^3F_2$ ,  $^3H$ , and  $^1I$ . If more data were available, a fitting starting with the parameters of table 17 could be undertaken.

The energy levels for  $\text{Ni}^{3+}$ ,  $\text{Cu}^{3+}$ , and  $\text{Zn}^{3+}$  were calculated using the parameters in table 17. We are unaware of any data reported on these ions in YAG, and we are uncertain as to the possibility of doping YAG with these ions. It would seem that doping with  $\text{Ni}^{3+}$  would be possible since the spectrum of this ion in the rare earth garnets has been reported [24].

**Author's note:** Since this report has been written an article has appeared (in Russian) by Yu. A. Voitukevich, M. V. Konzhik, V. V. Kuz'min, M. G. Livshitz and M. L. Meil'man: *Energy Structure of  $\text{Fe}^{3+}$  Impurity Ions in Yttrium Aluminum Garnet ( $\text{Y}_3\text{Al}_5\text{O}_{12}$ ) Crystals*, Opt. Spectrosc. 63 (1987), 810. Consequently, a supplement to this report will be issued which will include an analysis of this latest data.

## References

1. D. L. Wood and J. P. Remeika, *Optical Absorption of Tetrahedral  $\text{Co}^{3+}$  and  $\text{Co}^{2+}$  in Garnets*, J. Chem. Phys. 46 (1967), 3595.
2. C. A. Morrison, J. D. Bruno, and G. A. Turner, *Analysis of the Spectra of Triply Ionized Iron in Rare-Earth Aluminum Garnets*, Harry Diamond Laboratories, HDL-TR-2113 (September 1987).
3. E. Konig and S. Kremer, *Ligand Field Energy Diagrams*, Plenum Press, New York (1977).
4. I. I. Karpov, B. N. Grechusnikov, and Kh. S. Bagdarsarov, *Use of Polarized Luminescence to Investigate Luminescence in Yttrium-Aluminum Garnet Crystals with Added Titanium*, Sov. Phys. Crystallogr. 23 (1978), 688.
5. M. J. Weber and L. A. Riseberg, *Optical Spectra of Vanadium Ions in Yttrium Aluminum Garnet*, J. Chem. Phys. 55 (1971), 2032.
6. B. K. Sevast'yanov, D. T. Sviridov, V. P. Grekhovo, L. B. Pasternak, R. K. Sviridova, and T. F. Vermeichik, *Optical Absorption Spectra of Excited  $\text{Cr}^{3+}$  Ions in Yttrium Aluminum Garnet*, Sov. J. Electron. 2 (1973), 339.
7. D. T. Sviridov, R. K. Sviridova, N. I. Kulik, and V. B. Glasko, *Optical Spectra of the Iso-Electronic Ions  $\text{V}^{2+}$ ,  $\text{Cr}^{3+}$ , and  $\text{Mn}^{4+}$  in an Octahedral Coordination*, J. Appl. Spectrosc. 30 (1979), 334.
8. B. N. Grechusnikov, T. F. Veremeichik, and I. N. Kalinkina, *Analysis of Spectra of  $\text{Cr}^{3+}$  Impurity Ions in Series of Crystals Allowing for the Tree's Correction*, Sov. Phys. Crystallogr. 22 (1978), 525.
9. Z. T. Azamatov, P. A. Arsenev, T. Yu. Geraskina, and M. V. Chukichev, *Properties of Chromium Ions in the Lattice of Yttrium Aluminum Garnet (YAG)*, Phys. Status Solidi A: 1 (1970), 801.

10. M. O. Henry, J. P. Larkin, and G. F. Imbusch, *Luminescence from Chromium Doped Yttrium Aluminum Garnet*, Proceedings of the Royal Irish Academy, 75 (1975) 97.
11. P. A. Arsenov and D. T. Sviridov, *Absorption Spectra of Yttrium Aluminum Garnet (YAG) with Contaminant Ions of the Iron Group*, Sov. Phys. Crystallogr. 14 (1970), 578.
12. T. F. Veremeichik, B. N. Grechusnikov, T. M. Varina, D. T. Sviridov, and I. N. Kalinkina, *Absorption Spectra and Calculation of Energy-Level Diagram of  $Fe^{3+}$  and  $Mn^{2+}$  Ions in Single Crystals of Yttrium Aluminum Garnet, Orthoclase, and Manganese Silicate*, Sov. Phys. Crystallogr. 19 (1975), 742.
13. Y. Sugitani, K. Tagawa, and K. Kato, *Optical Absorption Spectra of Iron (III) and Chromium (III) Doped in Synthetic Yttrium-Aluminum-Garnets (YAG)*, Mineral J. 7 (1974), 445.
14. F. Euler and J. A. Bruce, *Oxygen Coordinates of Compounds with Garnet Structure*, Acta Cryst. 19 (1965), 971.
15. P. C. Schmidt, A. Weiss, and T. P. Das, *Effect of Crystal Fields and Self-Consistency on Dipole and Quadrupole Polarizabilities of Closed-Shell Ions*, Phys. Rev. B19 (1979), 5525.
16. N. Karayianis, C. A. Morrison, and D. E. Wortman, *Crystal Field Parameters for the Pentaphosphates of Neodymium and Europium*, J. Chem. Phys. 64 (1976), 3890.
17. R. P. Leavitt, *On the Rule of Certain Rotation Invariants in Crystal-Field Theory*, J. Chem. Phys. 77 (1982), 1661.
18. S. Fraga, K. M. S. Saxena, and J. Karwowski, *Physical Science Data: 5. Handbook of Atomic Data*, Elsevier, New York (1976). Also see C. A. Morrison and R. G. Schmalbach, *Approximate Values of  $\langle r^k \rangle$  for the Divalent, Trivalent, and Quadrivalent Ions with the  $3d^N$  Electronic Configuration*, Harry Diamond Laboratories, HDL-TL-85-3 (July 1985). Also see C. A. Morrison, N. Karayianis, and D. Wortman, *Rare-Earth Ion-Host Lattice Interactions 4. Predicting Spectra and Intensities of Lanthanides in Crystals*, Harry Diamond Laboratories, HDL-TR-1816 (June 1977).

19. R. M. Macfarlane, *Analysis of the Spectrum of  $d^3$  Ions in Trigonal Crystal Fields*, J. Chem. Phys. 39 (1963), 3118.
20. C. J. Ballhausen, *Ligand Field Theory*, McGraw-Hill, New York (1962), 104.
21. P. H. M. Uylings, A. J. J. Raassen, and J. F. Wyart, *Energies of  $N$  Equivalent Electrons Expressed in Terms of Two-Electron Energies and Independent Three-Electron Parameters: A New Complete Set of Orthogonal Operators: II. Applications of  $3d^N$  Configurations*, J. Phys. B17 (1984), 4103.
22. G. Burns, E. A. Geiss, B. A. Jenkins, and M. I. Nathan,  *$Cr^{3+}$  Fluorescence in Garnets and Other Crystals*, Phys. Rev. 139A (1965), 1687. (This reference gives the ground state splitting as  $0.53\text{ cm}^{-1}$  reported by J. W. Carson and R. L. White, J. Appl. Phys. 32 (1961), 1787. Also see L. J. Schruke and J. R. Cunningham, Jr., J. Appl. Phys. 37 (1966), 449.)
23. F. Bantien, P. Albers, and G. Huber, *Optical Transitions in Titanium-Doped YAG*, J. Lumin. 36 (1987), 363.
24. P. A. Arsenev, N. L. Perlova, and D. T. Sviridov, *Absorption Spectra of Nickel Ions in the Crystals of Some Rare Earth Aluminum Garnets*, Sov. Phys. Crystallogr. 16 (1971), 196.

# DISTRIBUTION

ADMINISTRATOR  
DEFENSE TECHNICAL INFORMATION CENTER  
ATTN DTIC-DDA (12 COPIES)  
CAMERON STATION, BUILDING 5  
ALEXANDRIA, VA 22314

DIRECTOR  
NIGHT VISION & ELECTRO-OPTICS LABORATORY  
ATTN TECHNICAL LIBRARY  
ATTN R. BUSER  
ATTN A. PINTO  
ATTN J. HABERSAT  
FT BELVOIR, VA 22060

DIRECTOR  
DEFENSE ADVANCED RESEARCH  
PROJECTS AGENCY  
ATTN J. FRIEBELE  
1400 WILSON BLVD  
ARLINGTON, VA 22209

DIRECTOR  
DEFENSE NUCLEAR AGENCY  
ATTN TECH LIBRARY  
WASHINGTON, DC 20305

UNDER SECRETARY OF DEFENSE RES  
& ENGINEERING  
ATTN TECHNICAL LIBRARY, 3C128  
WASHINGTON, DC 20301

OFFICE OF THE DEPUTY CHIEF OF STAFF,  
FOR RESEARCH, DEVELOPMENT,  
& ACQUISITION  
DEPARTMENT OF THE ARMY  
ATTN DAMA-ARZ-A, CHIEF SCIENTIST,  
L. CAMERON  
ATTN DAMA-ARZ-B, I. R. HERSHNER  
WASHINGTON, DC 20310

COMMANDER  
US ARMY ARMAMENT MUNITIONS &  
CHEMICAL COMMAND (AMCCOM)  
US ARMY ARMAMENT RESEARCH &  
DEVELOPMENT CENTER  
ATTN DRDAR-TSS, STINFO DIV  
DOVER, NJ 07801

COMMANDER  
ATMOSPHERIC SCIENCES LABORATORY  
ATTN TECHNICAL LIBRARY  
WHITE SANDS MISSILE RANGE, NM 88002

DIRECTOR  
US ARMY BALLISTIC RESEARCH LABORATORY  
ATTN DRDAR-TSB-S (STINFO)  
ABERDEEN PROVING GROUND, MD 21005

DIRECTOR  
US ARMY ELECTRONICS WARFARE LABORATORY  
ATTN J. CHARLTON  
ATTN DELET-DD  
FT MONMOUTH, NJ 07703

COMMANDING OFFICER  
USA FOREIGN SCIENCE & TECHNOLOGY CENTER  
FEDERAL OFFICE BUILDING  
ATTN DRXST-BS, BASIC SCIENCE DIV  
CHARLOTTESVILLE, VA 22901

COMMANDER  
US ARMY MATERIALS & MECHANICS  
RESEARCH CENTER  
ATTN DRXMR-TL, TECH LIBRARY  
WATERTOWN, MA 02172

US ARMY MATERIEL COMMAND  
5001 EISENHOWER AVE  
ALEXANDRIA, VA 22333-0001

US ARMY MATERIEL SYSTEMS ANALYSIS  
ACTIVITY  
ATTN DRXSY-MP (LIBRARY)  
ABERDEEN PROVING GROUND, MD 21005

COMMANDER  
US ARMY MISSILE & MUNITIONS  
CENTER & SCHOOL  
ATTN ATSK-CTD-F  
ATTN DRDMI-TB, REDSTONE SCI INFO CENTER  
REDSTONE ARSENAL, AL 35809

COMMANDER  
US ARMY RESEARCH OFFICE (DURHAM)  
PO BOX 12211  
ATTN ROBERT J. LONTZ  
ATTN M. STROSIO  
ATTN M. CIFTAN  
ATTN B. D. GUENTHER  
ATTN CHARLES BOGOSIAN  
RESEARCH TRIANGLE PARK, NC 27709

COMMANDER  
US ARMY RSCH & STD GRP (EUROPE)  
FPO NEW YORK 09510

COMMANDER  
US ARMY TEST & EVALUATION COMMAND  
ATTN D. H. SLINEY  
ATTN TECH LIBRARY  
ABERDEEN PROVING GROUND, MD 21005

DISTRIBUTION (cont'd)

COMMANDER  
US ARMY TROOP SUPPORT COMMAND  
ATTN DRXRES-RTL, TECH LIBRARY  
NATICK, MA 01762

OFFICE OF NAVAL RESEARCH  
ATTN J. MURDAY  
ARLINGTON, VA 22217

DIRECTOR  
NAVAL RESEARCH LABORATORY  
ATTN CODE 2620, TECH LIBRARY BR  
ATTN G. QUARLES  
ATTN G. KINTZ  
ATTN A. ROSENBAUM  
ATTN G. RISENBLATT  
ATTN CODE 5554, F. BARTOLI  
ATTN CODE 5554, L. ESTEROWITZ  
ATTN CODE 5554, R. E. ALLEN  
WASHINGTON, DC 20375

COMMANDER  
NAVAL WEAPONS CENTER  
ATTN CODE 3854, R. SCHWARTZ  
ATTN CODE 3854, M. HILLS  
ATTN CODE 3844, M. NADLER  
ATTN CODE 385, R. L. ATKINS  
ATTN CODE 343, TECHNICAL INFORMATION  
DEPARTMENT  
CHINA LAKE, CA 93555

AIR FORCE OFFICE OF SCIENTIFIC RESEARCH  
ATTN MAJOR H. V. WINSOR, USAF  
BOLLING AFB  
WASHINGTON, DC 20332

HQ, USAF/SAMI  
WASHINGTON, DC 20330

DEPARTMENT OF COMMERCE  
NATIONAL BUREAU OF STANDARDS  
ATTN LIBRARY  
WASHINGTON, DC 20234

NASA LANGLEY RESEARCH CENTER  
ATTN N. P. BARNES (2 COPIES)  
ATTN P. CROSS  
ATTN D. GETTENY  
ATTN J. BARNES  
ATTN C. BAIR  
HAMPTON, VA 23665

DIRECTOR  
ADVISORY GROUP ON ELECTRON DEVICES  
ATTN SECTRY, WORKING GROUP D  
201 VARICK STREET  
NEW YORK, NY 10013

AEROSPACE CORPORATION  
PO BOX 92957  
ATTN M. BIRNBAUM  
ATTN N. C. CHANG  
LOS ANGELES, CA 90009

COLORADO STATE UNIVERSITY  
PHYSICS DEPARTMENT  
ATTN S. KERN  
FORT COLLINS, CO 80523

ALLIED  
ADVANCED APPLICATION DEPT  
ATTN A. BUDGOR  
31717 LA TIEMDA DRIVE  
WESTLAKE VILLAGE, CA 91362

AMES LABORATORY DOE  
IOWA STATE UNIVERSITY  
ATTN K. A. GSCHNEIDNER, JR. (2 COPIES)  
AMES, IA 50011

ARGONNE NATIONAL LABORATORY  
ATTN W. T. CARNALL  
9700 SOUTH CASS AVENUE  
ARGONNE, IL 60439

ARIZONA STATE UNIVERSITY  
DEPT OF CHEMISTRY  
ATTN L. EYRING  
TEMPE, AZ 85281

BRIMROSE CORP OF AMERICA  
ATTN R. G. ROSEMEIER  
7527 BELAIR ROAD  
BALTIMORE, MD 21236

CARNEGIE MELLON UNIVERSITY  
SCHENLEY PARK  
ATTN PHYSICS & EE, J. O. ARTMAN  
PITTSBURGH, PA 15213

ENGINEERING SOCIETIES LIBRARY  
ATTN ACQUISITIONS DEPT  
345 EAST 47TH STREET  
NEW YORK, NY 10017

IBM RESEARCH DIVISION  
ALMADEN RESEARCH CENTER  
ATTN R. M. MACFARLANE  
MAIL STOP K32 802(D)  
650 HARRY ROAD  
SAN JOSE, CA 95120

# DISTRIBUTION

JOHNS HOPKINS UNIVERSITY  
DEPT OF PHYSICS  
ATTN B. R. JUDD  
BALTIMORE, MD 21218

KALAMAZOO COLLEGE  
DEPT OF PHYSICS  
ATTN K. RAJNAK  
KALAMAZOO, MI 49007

DIRECTOR  
LAWRENCE RADIATION LABORATORY  
ATTN MARVIN J. WEBER  
ATTN HELMUT A. KOEHLER  
ATTN W. KRUPKE  
LIVERMORE, CA 94550

MARTIN MARIETTA  
ATTN F. CROWNE  
ATTN R. LEAVITT  
ATTN J. LITTLE  
ATTN T. WORCHESKY  
ATTN D. WORTMAN  
1450 SOUTH ROLLING ROAD  
BALTIMORE, MD 21227

MASSACHUSETTS INSTITUTE OF TECHNOLOGY  
CRYSTAL PHYSICS LABORATORY  
ATTN H. P. JENSSSEN  
ATTN A. LINZ  
CAMBRIDGE, MA 02139

MASSACHUSETTS INSTITUTE OF TECHNOLOGY  
77 MASS AVE  
ROOM 26-251  
ATTN V. BAGNATO  
CAMBRIDGE, MA 02139

MIT LINCOLN LAB  
PO BOX 73  
ATTN PETER MOULTON,  
ATTN B. AULL  
LEXINGTON, MA 02173

DEPARTMENT OF MECHANICAL, INDUSTRIAL,  
& AEROSPACE ENGINEERING  
PO BOX 909  
ATTN S. TEMKIN  
PISCATAWAY, NJ 08854

NATIONAL OCEANIC & ATMOSPHERIC ADM  
ENVIRONMENTAL RESEARCH LABS  
ATTN LIBRARY, R-51, TECH RPTS  
BOULDER, CO 80302

OAK RIDGE NATIONAL LABORATORY  
ATTN R. G. HAIRE  
OAK RIDGE, TN 37830

SCIENCE APPLICATIONS, INC  
ATTN T. ALLIK  
1710 GOODRIDGE DRIVE  
McLEAN, VA 22102

UNION CARBIDE CORP  
ATTN M. R. KOKTA  
ATTN J. H. W. LIAW  
750 SOUTH 32ND STREET  
WASHOUGAL, WA 98671

UNIVERSITY OF MINNESOTA, DULUTH  
DEPARTMENT OF CHEMISTRY  
ATTN L. C. THOMPSON  
DULUTH, MN 55812

OKLAHOMA STATE UNIVERSITY  
DEPT OF PHYSICS  
ATTN R. C. POWELL  
STILLWATER, OK 74078

PENNSYLVANIA STATE UNIVERSITY  
MATERIALS RESEARCH LABORATORY  
ATTN W. B. WHITE  
ATTN B. K. CHANDRASEKHAR  
UNIVERSITY PARK, PA 16802

PRINCETON UNIVERSITY  
DEPARTMENT OF CHEMISTRY  
ATTN D. S. McCLURE  
PRINCETON, NJ 08544

SAN JOSE STATE UNIVERSITY  
DEPARTMENT OF PHYSICS  
ATTN J. B. GRUBER  
SAN JOSE, CA 95192

SETON HALL UNIVERSITY  
CHEMISTRY DEPARTMENT  
ATTN H. BRITTAIN  
SOUTH ORANGE, NJ 07099

UNIVERSITY OF VIRGINIA  
DEPT OF CHEMISTRY  
ATTN DR. F. S. RICHARDSON (2 COPIES)  
ATTN DR. M. REID  
CHARLOTTESVILLE, VA 22901

UNIVERSITY OF WISCONSIN  
CHEMISTRY DEPARTMENT  
ATTN J. WRIGHT  
ATTN B. TISSUE  
MADISON, WI 53706

US ARMY LABORATORY COMMAND  
ATTN TECHNICAL DIRECTOR, AMSLC-CT



DISTRIBUTION (cont'd)

INSTALLATION SUPPORT ACTIVITY  
ATTN LEGAL OFFICE, SLCIS-CC  
ATTN S. ELBAUM, SLCIS-CC

USAISC  
ATTN RECORD COPY, ASNC-LAB-TS  
ATTN TECHNICAL REPORTS BRANCH,  
ASNC-LAB-TR (2 COPIES)

HARRY DIAMOND LABORATORIES  
ATTN D/DIVISION DIRECTORS  
ATTN HDL LIBRARY, SLCHD-TL (3 COPIES)  
ATTN HDL LIBRARY, SLCHD-TL (WOODBRIDGE)  
ATTN CHIEF, SLCHD-NW-E  
ATTN CHIEF, SLCHD-NW-EP  
ATTN CHIEF, SLCHD-NW-EH  
ATTN CHIEF, SLCHD-NW-ES  
ATTN CHIEF, SLCHD-NW-R  
ATTN CHIEF, SLCHD-NW-TN  
ATTN CHIEF, SLCHD-NW-RP  
ATTN CHIEF, SLCHD-NW-CS  
ATTN CHIEF, SLCHD-NW-TS  
ATTN CHIEF, SLCHD-NW-RS  
ATTN CHIEF, SLCHD-NW-P  
ATTN CHIEF, SLCHD-PO  
ATTN CHIEF, SLCHD-ST-C  
ATTN CHIEF, SLCHD-ST-RS  
ATTN CHIEF, SLCHD-ST-RP  
ATTN CHIEF, SLCHD-TT  
ATTN WILLIS, B., SLCHD-TA-ET  
ATTN ZABLUDOWSKI, B., SLCHD-TA-ET  
ATTN KENYON, C. S., SLCHD-NW-EP

HARRY DIAMOND LABORATORIES (cont'd)  
ATTN MILETTA J. R., SLCHD-NW-EP  
ATTN MCLEAN, F. B., SLCHD-NW-RP  
ATTN LIBELO, L., SLCHD-ST-MW  
ATTN BENCIVENGA, A. A., SLCHD-ST-SP  
ATTN SATTTLER, J., SLCHD-PO-P  
ATTN NEMARICH, J., SLCHD-ST-CB  
ATTN WEBER, B., SLCHD-ST-CB  
ATTN BAHDER, T., SLCHD-ST-AP  
ATTN BENCIVENGA, B. SLCHD-ST-AP  
ATTN BRODY, P., SLCHD-ST-AP  
ATTN BRUNO, J., SLCHD-ST-AP  
ATTN DROPKIN, H., SLCHD-ST-AP  
ATTN EDWARDS, A., SLCHD-ST-AP  
ATTN HANSEN, A., SLCHD-ST-AP  
ATTN HAY, G., SLCHD-ST-AP  
ATTN KATZEN, E., SLCHD-ST-AP  
ATTN NEIFELD, R., SLCHD-ST-AP  
ATTN PENNISE, C., SLCHD-ST-AP  
ATTN SCHMALBACH, R., SLCHD-ST-AP  
ATTN SEMENDY, F., SLCHD-ST-AP  
ATTN SIMONIS, G., SLCHD-ST-AP  
ATTN SIMPSON, T., SLCHD-ST-AP  
ATTN STEAD, M., SLCHD-ST-AP  
ATTN STELLATO, J., SLCHD-ST-AP  
ATTN TOBIN, M., SLCHD-ST-AP  
ATTN TURNER, G., SLCHD-ST-AP (10 COPIES)  
ATTN WONG, B., SLCHD-ST-AP  
ATTN WORTMAN, D., SLCHD-ST-AP  
ATTN GARVIN, C., SLCHD-ST-SS  
ATTN GOFF, J., SLCHD-ST-SS  
ATTN MORRISON, C., SLCHD-ST-AP (10 COPIES)